

Acetoxyacetic acid, 2-(1-adamantyl)ethyl ester

Inchi:	InChI=1S/C16H24O4/c1-11(17)20-10-15(18)19-3-2-16-7-12-4-13(8-16)6-14(5-12)9-16/h1
InchiKey:	ANKZCVBEWNJHPO-UHFFFAOYSA-N
Formula:	C16H24O4
SMILES:	CC(=O)OCC(=O)OCCC12CC3CC(CC(C3)C1)C2
Mol. weight [g/mol]:	280.36

Physical Properties

Property code	Value	Unit	Source
gf	-227.05	kJ/mol	Joback Method
hf	-656.03	kJ/mol	Joback Method
hfus	29.85	kJ/mol	Joback Method
hvap	67.97	kJ/mol	Joback Method
log10ws	-2.96		Crippen Method
logp	2.699		Crippen Method
mcvol	218.600	ml/mol	McGowan Method
pc	2001.91	kPa	Joback Method
rinsol	2009.00		NIST Webbook
tb	738.12	K	Joback Method
tc	950.84	K	Joback Method
tf	484.36	K	Joback Method
vc	0.840	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	689.70	J/mol×K	738.12	Joback Method
cpg	708.23	J/mol×K	773.57	Joback Method
cpg	725.98	J/mol×K	809.03	Joback Method
cpg	743.10	J/mol×K	844.48	Joback Method
cpg	759.76	J/mol×K	879.93	Joback Method
cpg	776.10	J/mol×K	915.38	Joback Method
cpg	792.30	J/mol×K	950.84	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U308317&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mccvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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