

# Glutaric acid, isohexyl 2,4,5-trichlorophenyl ester

<b>Inchi:</b>	InChI=1S/C17H21Cl3O4/c1-11(2)5-4-8-23-16(21)6-3-7-17(22)24-15-10-13(19)12(18)9-14
<b>InchiKey:</b>	IDVQAXGPMBGICG-UHFFFAOYSA-N
<b>Formula:</b>	C17H21Cl3O4
<b>SMILES:</b>	CC(C)CCCOC(=O)CCCC(=O)Oc1cc(Cl)c(Cl)cc1Cl
<b>Mol. weight [g/mol]:</b>	395.70

## Physical Properties

Property code	Value	Unit	Source
gf	-330.29	kJ/mol	Joback Method
hf	-734.19	kJ/mol	Joback Method
hfus	47.30	kJ/mol	Joback Method
hvap	88.78	kJ/mol	Joback Method
log10ws	-6.23		Crippen Method
logp	5.702		Crippen Method
mcvol	278.230	ml/mol	McGowan Method
pc	1508.15	kPa	Joback Method
rinpola	2666.00		NIST Webbook
tb	894.41	K	Joback Method
tc	1111.41	K	Joback Method
tf	564.41	K	Joback Method
vc	1.069	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	777.49	J/molxK	894.41	Joback Method
cpg	789.38	J/molxK	930.58	Joback Method
cpg	800.17	J/molxK	966.74	Joback Method
cpg	809.86	J/molxK	1002.91	Joback Method
cpg	818.46	J/molxK	1039.08	Joback Method
cpg	826.00	J/molxK	1075.25	Joback Method
cpg	832.48	J/molxK	1111.41	Joback Method
dvisc	0.0003779	Paxs	564.41	Joback Method
dvisc	0.0002328	Paxs	619.41	Joback Method

dvisc	0.0001552	Paxs	674.41	Joback Method
dvisc	0.0001100	Paxs	729.41	Joback Method
dvisc	0.0000818	Paxs	784.41	Joback Method
dvisc	0.0000633	Paxs	839.41	Joback Method
dvisc	0.0000505	Paxs	894.41	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U359107&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U359107&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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