

# 1,4-Dioxane, 2-ethyl-5-methyl-

<b>Other names:</b>	2-Ethyl-5-methyl-1,4-dioxane
<b>Inchi:</b>	InChI=1S/C7H14O2/c1-3-7-5-8-6(2)4-9-7/h6-7H,3-5H2,1-2H3
<b>InchiKey:</b>	DPXGZEPGOMVZDJ-UHFFFAOYSA-N
<b>Formula:</b>	C7H14O2
<b>SMILES:</b>	CCC1COC(C)CO1
<b>Mol. weight [g/mol]:</b>	130.18
<b>CAS:</b>	53907-91-8

## Physical Properties

Property code	Value	Unit	Source
gf	-147.44	kJ/mol	Joback Method
hf	-417.83	kJ/mol	Joback Method
hfus	22.75	kJ/mol	Joback Method
hvap	40.32	kJ/mol	Joback Method
log10ws	-1.05		Crippen Method
logp	1.200		Crippen Method
mcvol	110.370	ml/mol	McGowan Method
pc	3318.18	kPa	Joback Method
rinpol	913.00		NIST Webbook
tb	428.34	K	Joback Method
tc	630.96	K	Joback Method
tf	224.93	K	Joback Method
vc	0.402	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	234.11	J/molxK	428.34	Joback Method
cpg	305.59	J/molxK	597.19	Joback Method
cpg	292.65	J/molxK	563.42	Joback Method
cpg	279.04	J/molxK	529.65	Joback Method
cpg	264.75	J/molxK	495.88	Joback Method
cpg	249.78	J/molxK	462.11	Joback Method
cpg	317.87	J/molxK	630.96	Joback Method

dvisc	0.0003579	Paxs	428.34	Joback Method
dvisc	0.0004645	Paxs	394.44	Joback Method
dvisc	0.0006333	Paxs	360.54	Joback Method
dvisc	0.0009207	Paxs	326.63	Joback Method
dvisc	0.0014597	Paxs	292.73	Joback Method
dvisc	0.0026113	Paxs	258.83	Joback Method
dvisc	0.0055665	Paxs	224.93	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C53907918&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C53907918&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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