

# 2-(2,4,5-Trichlorophenoxy)ethanol

<b>Other names:</b>	Ethanol, 2-(2,4,5-trichlorophenoxy)-
<b>Inchi:</b>	InChI=1S/C8H7Cl3O2/c9-5-3-7(11)8(4-6(5)10)13-2-1-12/h3-4,12H,1-2H2
<b>InchiKey:</b>	ATKFMEGWDYLXBP-UHFFFAOYSA-N
<b>Formula:</b>	C8H7Cl3O2
<b>SMILES:</b>	OCCOc1cc(Cl)c(Cl)cc1Cl
<b>Mol. weight [g/mol]:</b>	241.50
<b>CAS:</b>	2122-77-2

## Physical Properties

Property code	Value	Unit	Source
gf	-177.61	kJ/mol	Joback Method
hf	-338.00	kJ/mol	Joback Method
hfus	27.22	kJ/mol	Joback Method
hvap	69.91	kJ/mol	Joback Method
log10ws	-3.33		Crippen Method
logp	3.018		Crippen Method
mcvol	148.280	ml/mol	McGowan Method
pc	3318.18	kPa	Joback Method
tb	650.95	K	Joback Method
tc	862.24	K	Joback Method
tf	416.71	K	Joback Method
vc	0.559	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	308.52	J/molxK	650.95	Joback Method
cpg	343.21	J/molxK	827.02	Joback Method
cpg	337.18	J/molxK	791.81	Joback Method
cpg	330.70	J/molxK	756.59	Joback Method
cpg	323.77	J/molxK	721.38	Joback Method
cpg	316.37	J/molxK	686.16	Joback Method
cpg	348.78	J/molxK	862.24	Joback Method
dvisc	0.0000621	Paxs	650.95	Joback Method

dvisc	0.0000859	Paxs	611.91	Joback Method
dvisc	0.0001240	Paxs	572.87	Joback Method
dvisc	0.0001889	Paxs	533.83	Joback Method
dvisc	0.0003076	Paxs	494.79	Joback Method
dvisc	0.0005446	Paxs	455.75	Joback Method
dvisc	0.0010729	Paxs	416.71	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2122772&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2122772&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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