

# Phthalic acid, decyl 3-fluorobenzyl ester

<b>Inchi:</b>	InChI=1S/C25H31FO4/c1-2-3-4-5-6-7-8-11-17-29-24(27)22-15-9-10-16-23(22)25(28)30-
<b>InchiKey:</b>	XAIZCEQRRYICHN-UHFFFAOYSA-N
<b>Formula:</b>	C25H31FO4
<b>SMILES:</b>	CCCCCCCCCOC(=O)c1ccccc1C(=O)OCc1ccc(F)c1
<b>Mol. weight [g/mol]:</b>	414.51

## Physical Properties

Property code	Value	Unit	Source
gf	-297.47	kJ/mol	Joback Method
hf	-794.92	kJ/mol	Joback Method
hfus	56.46	kJ/mol	Joback Method
hvap	94.61	kJ/mol	Joback Method
log10ws	-8.17		Crippen Method
logp	6.480		Crippen Method
mcvol	332.240	ml/mol	McGowan Method
pc	1157.71	kPa	Joback Method
rinsol	2951.00		NIST Webbook
tb	986.57	K	Joback Method
tc	1209.75	K	Joback Method
tf	594.30	K	Joback Method
vc	1.286	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1082.82	J/mol×K	986.57	Joback Method
cpg	1096.93	J/mol×K	1023.77	Joback Method
cpg	1109.61	J/mol×K	1060.96	Joback Method
cpg	1120.93	J/mol×K	1098.16	Joback Method
cpg	1130.93	J/mol×K	1135.35	Joback Method
cpg	1139.66	J/mol×K	1172.55	Joback Method
cpg	1147.18	J/mol×K	1209.75	Joback Method

# Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U377894&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U377894&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h vap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m cvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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