

Butane, 2-iodo-

Other names:	2-Butyl Iodide 2-IODOBUTANE 2-Jodbutan SEC-BUTYL IODIDE SEC-IODOBUTANE UN 2390 sec-C4H9I
Inchi:	InChI=1S/C4H9I/c1-3-4(2)5/h4H,3H2,1-2H3
InchiKey:	IQRUSQUYPCKEKN-UHFFFAOYSA-N
Formula:	C4H9I
SMILES:	CCC(C)I
Mol. weight [g/mol]:	184.02
CAS:	513-48-4

Physical Properties

Property code	Value	Unit	Source
gf	38.48	kJ/mol	Joback Method
hf	-54.30	kJ/mol	Joback Method
hfus	7.00	kJ/mol	Joback Method
hvap	37.90	kJ/mol	NIST Webbook
hvap	38.49	kJ/mol	NIST Webbook
hvap	38.46 ± 0.06	kJ/mol	NIST Webbook
hvap	38.50 ± 0.10	kJ/mol	NIST Webbook
hvap	38.80	kJ/mol	NIST Webbook
ie	9.40	eV	NIST Webbook
ie	9.09 ± 0.02	eV	NIST Webbook
ie	9.13	eV	NIST Webbook
ie	9.10	eV	NIST Webbook
ie	9.10	eV	NIST Webbook
ie	9.10 ± 0.02	eV	NIST Webbook
log10ws	-2.56		Crippen Method
logp	2.220		Crippen Method
mcvol	93.040	ml/mol	McGowan Method
pc	3829.28	kPa	Joback Method
rinpol	789.00		NIST Webbook
ripol	1019.00		NIST Webbook
ripol	1019.00		NIST Webbook

tb	392.70 ± 2.00	K	NIST Webbook
tb	393.00	K	NIST Webbook
tb	393.20	K	NIST Webbook
tb	392.70	K	NIST Webbook
tb	391.40 ± 1.00	K	NIST Webbook
tb	391.15 ± 1.50	K	NIST Webbook
tb	393.15	K	KDB
tc	593.02	K	Joback Method
tf	168.95 ± 0.40	K	NIST Webbook
tf	133.15	K	KDB
vc	0.342	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	137.04	J/mol×K	383.62	Joback Method
cpg	145.49	J/mol×K	418.52	Joback Method
cpg	153.49	J/mol×K	453.42	Joback Method
cpg	161.06	J/mol×K	488.32	Joback Method
cpg	168.23	J/mol×K	523.22	Joback Method
cpg	175.01	J/mol×K	558.12	Joback Method
cpg	181.41	J/mol×K	593.02	Joback Method
cpl	165.30	J/mol×K	298.15	NIST Webbook
dvisc	0.0101698	Pa×s	177.90	Joback Method
dvisc	0.0038120	Pa×s	212.19	Joback Method
dvisc	0.0018774	Pa×s	246.47	Joback Method
dvisc	0.0010993	Pa×s	280.76	Joback Method
dvisc	0.0007232	Pa×s	315.05	Joback Method
dvisc	0.0005165	Pa×s	349.33	Joback Method
dvisc	0.0003918	Pa×s	383.62	Joback Method
hvapt	33.27	kJ/mol	393.20	NIST Webbook

Correlations

Information	Value
Property code	pvap
Equation	ln(Pvp) = A + B/(T + C)
Coeff. A	1.76031e+01

Coeff. B	-5.19995e+03
Coeff. C	8.46400e+00
Temperature range (K), min.	291.84
Temperature range (K), max.	414.58

Sources

KDB:	https://www.thermo.com/research/kdb/hcprop/showprop.php?cmpid=1612
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C513484&Units=SI
The Yaws Handbook of Vapor Pressure:	https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemed.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
cpl:	Liquid phase heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
hvapt:	Enthalpy of vaporization at a given temperature
ie:	Ionization energy
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
pvap:	Vapor pressure
rinpol:	Non-polar retention indices
ripol:	Polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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