

# di-(3-Methoxybutyl)succinate

<b>Inchi:</b>	InChI=1S/C14H26O6/c1-11(17-3)7-9-19-13(15)5-6-14(16)20-10-8-12(2)18-4/h11-12H,5-
<b>InchiKey:</b>	NQQXSMSMEDVFBP-UHFFFAOYSA-N
<b>Formula:</b>	C14H26O6
<b>SMILES:</b>	<chem>COC(C)CCOC(=O)CCC(=O)OCCC(C)OC</chem>
<b>Mol. weight [g/mol]:</b>	290.35

## Physical Properties

Property code	Value	Unit	Source
gf	-615.72	kJ/mol	Joback Method
hf	-1096.89	kJ/mol	Joback Method
hfus	32.92	kJ/mol	Joback Method
hvap	69.11	kJ/mol	Joback Method
log10ws	-1.81		Crippen Method
logp	1.703		Crippen Method
mvol	234.740	ml/mol	McGowan Method
pc	1615.47	kPa	Joback Method
rinpol	1829.00		NIST Webbook
tb	716.26	K	Joback Method
tc	897.43	K	Joback Method
tf	406.32	K	Joback Method
vc	0.891	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	680.91	J/molxK	716.26	Joback Method
cpg	696.60	J/molxK	746.46	Joback Method
cpg	711.47	J/molxK	776.65	Joback Method
cpg	725.50	J/molxK	806.85	Joback Method
cpg	738.69	J/molxK	837.04	Joback Method
cpg	751.03	J/molxK	867.24	Joback Method
cpg	762.49	J/molxK	897.43	Joback Method
dvisc	0.0010461	Paxs	406.32	Joback Method
dvisc	0.0004965	Paxs	457.98	Joback Method

dvisc	0.0002740	Paxs	509.63	Joback Method
dvisc	0.0001687	Paxs	561.29	Joback Method
dvisc	0.0001128	Paxs	612.95	Joback Method
dvisc	0.0000802	Paxs	664.60	Joback Method
dvisc	0.0000599	Paxs	716.26	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R542288&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R542288&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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