

# Benzene, 1-bromo-2-(chloromethyl)-

<b>Inchi:</b>	InChI=1S/C7H6BrCl/c8-7-4-2-1-3-6(7)5-9/h1-4H,5H2
<b>InchiKey:</b>	DDVSFIUKWUTKES-UHFFFAOYSA-N
<b>Formula:</b>	C7H6BrCl
<b>SMILES:</b>	ClCc1ccccc1Br
<b>Mol. weight [g/mol]:</b>	205.48
<b>CAS:</b>	578-51-8

## Physical Properties

Property code	Value	Unit	Source
gf	113.23	kJ/mol	Joback Method
hf	47.84	kJ/mol	Joback Method
hfus	17.02	kJ/mol	Joback Method
hvap	44.93	kJ/mol	Joback Method
log10ws	-3.67		Crippen Method
logp	3.188		Crippen Method
mcvol	115.470	ml/mol	McGowan Method
pc	4205.63	kPa	Joback Method
rinpol	1244.00		NIST Webbook
rinpol	1244.00		NIST Webbook
tb	494.81	K	Joback Method
tc	734.55	K	Joback Method
tf	297.31	K	Joback Method
vc	0.430	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	195.64	J/molxK	494.81	Joback Method
cpg	237.24	J/molxK	694.60	Joback Method
cpg	230.19	J/molxK	654.64	Joback Method
cpg	222.55	J/molxK	614.68	Joback Method
cpg	214.28	J/molxK	574.72	Joback Method
cpg	205.32	J/molxK	534.77	Joback Method
cpg	243.74	J/molxK	734.55	Joback Method

dvisc	0.0003139	Paxs	494.81	Joback Method
dvisc	0.0003844	Paxs	461.89	Joback Method
dvisc	0.0004857	Paxs	428.98	Joback Method
dvisc	0.0006381	Paxs	396.06	Joback Method
dvisc	0.0008806	Paxs	363.14	Joback Method
dvisc	0.0012960	Paxs	330.23	Joback Method
dvisc	0.0020778	Paxs	297.31	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C578518&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C578518&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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