

# Diethylmalonic acid, monochloride, decyl ester

Inchi:	InChI=1S/C17H31ClO3/c1-4-7-8-9-10-11-12-13-14-21-16(20)17(5-2,6-3)15(18)19/h4-14H
InchiKey:	KSWCHIVJMVMIU-UHFFFAOYSA-N
Formula:	C17H31ClO3
SMILES:	CCCCCCCCCOC(=O)C(CC)(CC)C(=O)Cl
Mol. weight [g/mol]:	318.88

## Physical Properties

Property code	Value	Unit	Source
gf	-279.67	kJ/mol	Joback Method
hf	-776.08	kJ/mol	Joback Method
hfus	40.95	kJ/mol	Joback Method
hvap	72.43	kJ/mol	Joback Method
log10ws	-5.49		Crippen Method
logp	5.242		Crippen Method
mvol	271.640	ml/mol	McGowan Method
pc	1313.70	kPa	Joback Method
rinpol	1928.00		NIST Webbook
rinpol	1928.00		NIST Webbook
tb	752.72	K	Joback Method
tc	938.73	K	Joback Method
tf	435.78	K	Joback Method
vc	1.056	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	794.32	J/molxK	752.72	Joback Method
cpg	868.16	J/molxK	907.73	Joback Method
cpg	855.08	J/molxK	876.73	Joback Method
cpg	841.19	J/molxK	845.73	Joback Method
cpg	826.46	J/molxK	814.72	Joback Method
cpg	810.85	J/molxK	783.72	Joback Method
cpg	880.47	J/molxK	938.73	Joback Method
dvisc	0.0000768	Paxs	752.72	Joback Method

dvisc	0.0001032	Paxs	699.90	Joback Method
dvisc	0.0001453	Paxs	647.07	Joback Method
dvisc	0.0002176	Paxs	594.25	Joback Method
dvisc	0.0003524	Paxs	541.43	Joback Method
dvisc	0.0006336	Paxs	488.60	Joback Method
dvisc	0.0013130	Paxs	435.78	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U369507&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U369507&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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