

# Cyclopentane, methylene-

<b>Other names:</b>	Methylenecyclopentane 1-Methylenecyclopentane
<b>Inchi:</b>	InChI=1S/C6H10/c1-6-4-2-3-5-6/h1-5H2
<b>InchiKey:</b>	NFJPEKRRHIYYES-UHFFFAOYSA-N
<b>Formula:</b>	C6H10
<b>SMILES:</b>	C=C1CCCC1
<b>Mol. weight [g/mol]:</b>	82.14
<b>CAS:</b>	1528-30-9

## Physical Properties

Property code	Value	Unit	Source
affp	832.40	kJ/mol	NIST Webbook
basg	803.50	kJ/mol	NIST Webbook
chl	-3770.10 ± 0.79	kJ/mol	NIST Webbook
gf	96.98	kJ/mol	Joback Method
hf	10.20	kJ/mol	NIST Webbook
hfl	-20.20 ± 0.88	kJ/mol	NIST Webbook
hfus	3.00	kJ/mol	Joback Method
hvap	29.68	kJ/mol	Joback Method
ie	8.70 ± 0.05	eV	NIST Webbook
ie	9.05 ± 0.02	eV	NIST Webbook
ie	9.14	eV	NIST Webbook
ie	9.26 ± 0.05	eV	NIST Webbook
ie	8.55 ± 0.01	eV	NIST Webbook
ie	8.51 ± 0.01	eV	NIST Webbook
ie	8.94 ± 0.01	eV	NIST Webbook
ie	8.96 ± 0.02	eV	NIST Webbook
ie	7.20	eV	NIST Webbook
log10ws	-2.08		Crippen Method
logp	2.117		Crippen Method
mcvol	80.240	ml/mol	McGowan Method
pc	4098.62	kPa	Joback Method
rinpol	650.00		NIST Webbook
rinpol	654.00		NIST Webbook
rinpol	657.00		NIST Webbook
rinpol	650.00		NIST Webbook
rinpol	657.00		NIST Webbook

rmpol	654.00		NIST Webbook
rmpol	654.00		NIST Webbook
rmpol	657.00		NIST Webbook
rmpol	657.00		NIST Webbook
rmpol	654.00		NIST Webbook
tb	355.79	K	Joback Method
tc	553.33	K	Joback Method
tf	146.00 ± 0.15	K	NIST Webbook
tf	145.32 ± 0.50	K	NIST Webbook
vc	0.297	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	126.63	J/mol×K	355.79	Joback Method
cpg	138.59	J/mol×K	388.71	Joback Method
cpg	149.93	J/mol×K	421.64	Joback Method
cpg	160.70	J/mol×K	454.56	Joback Method
cpg	170.91	J/mol×K	487.48	Joback Method
cpg	180.58	J/mol×K	520.40	Joback Method
cpg	189.73	J/mol×K	553.33	Joback Method
dvisc	0.0028005	Paxs	186.20	Joback Method
dvisc	0.0015005	Paxs	214.46	Joback Method
dvisc	0.0009297	Paxs	242.73	Joback Method
dvisc	0.0006366	Paxs	271.00	Joback Method
dvisc	0.0004682	Paxs	299.26	Joback Method
dvisc	0.0003631	Paxs	327.52	Joback Method
dvisc	0.0002932	Paxs	355.79	Joback Method

## Sources

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C1528309&Units=SI>

# Legend

<b>affp:</b>	Proton affinity
<b>basg:</b>	Gas basicity
<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpolar:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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