

# Diethylmalonic acid, ethyl 3-methylbenzyl ester

Inchi:	InChI=1S/C17H24O4/c1-5-17(6-2,15(18)20-7-3)16(19)21-12-14-10-8-9-13(4)11-14/h8-11
InchiKey:	PFTLVKPNEYWMPF-UHFFFAOYSA-N
Formula:	C17H24O4
SMILES:	CCOC(=O)C(CC)(CC)C(=O)OCc1cccc(C)c1
Mol. weight [g/mol]:	292.37

## Physical Properties

Property code	Value	Unit	Source
gf	-269.96	kJ/mol	Joback Method
hf	-667.50	kJ/mol	Joback Method
hfus	31.60	kJ/mol	Joback Method
hvap	73.39	kJ/mol	Joback Method
log10ws	-4.08		Crippen Method
logp	3.408		Crippen Method
mcvol	241.510	ml/mol	McGowan Method
pc	1703.31	kPa	Joback Method
rinpol	1891.00		NIST Webbook
rinpol	1891.00		NIST Webbook
tb	769.37	K	Joback Method
tc	977.67	K	Joback Method
tf	467.03	K	Joback Method
vc	0.916	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	706.20	J/molxK	769.37	Joback Method
cpg	721.92	J/molxK	804.09	Joback Method
cpg	736.56	J/molxK	838.80	Joback Method
cpg	750.15	J/molxK	873.52	Joback Method
cpg	762.72	J/molxK	908.23	Joback Method
cpg	774.31	J/molxK	942.95	Joback Method
cpg	784.95	J/molxK	977.67	Joback Method
dvisc	0.0007474	Paxs	467.03	Joback Method

dvisc	0.0004108	Paxs	517.42	Joback Method
dvisc	0.0002511	Paxs	567.81	Joback Method
dvisc	0.0001663	Paxs	618.20	Joback Method
dvisc	0.0001172	Paxs	668.59	Joback Method
dvisc	0.0000867	Paxs	718.98	Joback Method
dvisc	0.0000668	Paxs	769.37	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U369303&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U369303&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>c</sub>vol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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