

# 2-Ethylbutyric acid, dodecyl ester

**Inchi:** InChI=1S/C18H36O2/c1-4-7-8-9-10-11-12-13-14-15-16-20-18(19)17(5-2)6-3/h17H,4-16H  
**InchiKey:** NOZDRKNOMKZEFU-UHFFFAOYSA-N  
**Formula:** C18H36O2  
**SMILES:** CCCCCCCCCCOC(=O)C(CC)CC  
**Mol. weight [g/mol]:** 284.48

## Physical Properties

Property code	Value	Unit	Source
gf	-135.68	kJ/mol	Joback Method
hf	-664.93	kJ/mol	Joback Method
hfus	41.64	kJ/mol	Joback Method
hvap	64.43	kJ/mol	Joback Method
log10ws	-5.98		Crippen Method
logp	5.887		Crippen Method
mcvol	271.920	ml/mol	McGowan Method
pc	1195.65	kPa	Joback Method
rinpola	1874.00		NIST Webbook
rinpola	1874.00		NIST Webbook
rinpola	1915.00		NIST Webbook
rinpola	1915.00		NIST Webbook
tb	687.09	K	Joback Method
tc	857.14	K	Joback Method
tf	349.78	K	Joback Method
vc	1.062	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	790.33	J/molxK	687.09	Joback Method
cpg	877.02	J/molxK	828.80	Joback Method
cpg	861.31	J/molxK	800.46	Joback Method
cpg	844.81	J/molxK	772.12	Joback Method
cpg	827.49	J/molxK	743.77	Joback Method
cpg	809.34	J/molxK	715.43	Joback Method

cpg	891.94	J/mol×K	857.14	Joback Method
dvisc	0.0000883	Paxs	687.09	Joback Method
dvisc	0.0001207	Paxs	630.87	Joback Method
dvisc	0.0001755	Paxs	574.65	Joback Method
dvisc	0.0002766	Paxs	518.43	Joback Method
dvisc	0.0004870	Paxs	462.22	Joback Method
dvisc	0.0010029	Paxs	406.00	Joback Method
dvisc	0.0026053	Paxs	349.78	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U369723&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U369723&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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