

# 1,2-Cyclopentanedione, 3-methyl-

<b>Other names:</b>	3-Methyl-1,2-cyclopentanedione 3-Methylcyclopentane-1,2-dione
<b>Inchi:</b>	InChI=1S/C6H8O2/c1-4-2-3-5(7)6(4)8/h4H,2-3H2,1H3
<b>InchiKey:</b>	OACYKCIZDVVNJL-UHFFFAOYSA-N
<b>Formula:</b>	C6H8O2
<b>SMILES:</b>	CC1CCC(=O)C1=O
<b>Mol. weight [g/mol]:</b>	112.13
<b>CAS:</b>	765-70-8

## Physical Properties

Property code	Value	Unit	Source
gf	-208.99	kJ/mol	Joback Method
hf	-382.09	kJ/mol	Joback Method
hfus	4.25	kJ/mol	Joback Method
hvap	37.70	kJ/mol	Joback Method
log10ws	-0.55		Crippen Method
logp	0.554		Crippen Method
mcvol	87.680	ml/mol	McGowan Method
pc	4194.74	kPa	Joback Method
rinpol	1043.00		NIST Webbook
rinpol	1043.00		NIST Webbook
rinpol	1043.00		NIST Webbook
ripol	1781.00		NIST Webbook
tb	487.60	K	Joback Method
tc	724.67	K	Joback Method
tf	304.72	K	Joback Method
vc	0.327	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	188.84	J/molxK	487.60	Joback Method
cpg	201.89	J/molxK	527.11	Joback Method
cpg	214.49	J/molxK	566.62	Joback Method

cpg	226.61	J/mol×K	606.13	Joback Method
cpg	238.19	J/mol×K	645.64	Joback Method
cpg	249.18	J/mol×K	685.15	Joback Method
cpg	259.55	J/mol×K	724.67	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C765708&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C765708&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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