

3,4-dihydro-2H-thiopyran-3-one

Inchi:	InChI=1S/C5H6OS/c7-5-2-1-3-6-4-5/h1,3H,2,4H2
InchiKey:	FUTCFJQVPUDGAC-UHFFFAOYSA-N
Formula:	C5H6OS
SMILES:	S=C1CC=CO1
Mol. weight [g/mol]:	114.17

Physical Properties

Property code	Value	Unit	Source
gf	58.07	kJ/mol	Joback Method
hf	-30.99	kJ/mol	Joback Method
hfus	14.71	kJ/mol	Joback Method
hvap	39.74	kJ/mol	Joback Method
log10ws	-1.60		Crippen Method
logp	1.290		Crippen Method
mcvol	84.070	ml/mol	McGowan Method
pc	5335.72	kPa	Joback Method
rinpol	1155.00		NIST Webbook
tb	436.77	K	Joback Method
tc	673.38	K	Joback Method
tf	248.73	K	Joback Method
vc	0.294	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	147.37	J/mol×K	436.77	Joback Method
cpg	157.36	J/mol×K	476.20	Joback Method
cpg	166.57	J/mol×K	515.64	Joback Method
cpg	175.04	J/mol×K	555.07	Joback Method
cpg	182.84	J/mol×K	594.51	Joback Method
cpg	190.00	J/mol×K	633.94	Joback Method
cpg	196.59	J/mol×K	673.38	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R223833&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mccvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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