

# 2-Pentanol, 2-methyl-

<b>Other names:</b>	1,1-Dimethylbutanol 2-Hydroxy-2-methylpentane 2-Methyl-2-hydroxypentane 2-Methyl-2-pentanol 2-Methylpentan-2-ol Methyl-2 pentanol-2 UN 2560 dimethylpropylcarbinol
<b>Inchi:</b>	InChI=1S/C6H14O/c1-4-5-6(2,3)7/h7H,4-5H2,1-3H3
<b>InchiKey:</b>	WFRBDWRZVBPBDO-UHFFFAOYSA-N
<b>Formula:</b>	C6H14O
<b>SMILES:</b>	CCCC(C)(C)O
<b>Mol. weight [g/mol]:</b>	102.17
<b>CAS:</b>	590-36-3

## Physical Properties

Property code	Value	Unit	Source
gf	-134.34	kJ/mol	Joback Method
hf	-328.15	kJ/mol	Joback Method
hfus	7.97	kJ/mol	Joback Method
hvap	54.82	kJ/mol	NIST Webbook
h <sub>vap</sub>	54.70 ± 0.20	kJ/mol	NIST Webbook
log <sub>10</sub> ws	-0.49		Estimated Solubility Method
log <sub>10</sub> ws	-0.49		Aqueous Solubility Prediction Method
logp	1.557		Crippen Method
m <sub>cvol</sub>	101.270	ml/mol	McGowan Method
pc	3530.46	kPa	Joback Method
r <sub>inpol</sub>	718.00		NIST Webbook
r <sub>inpol</sub>	719.00		NIST Webbook
r <sub>inpol</sub>	718.00		NIST Webbook
r <sub>inpol</sub>	718.00		NIST Webbook
r <sub>inpol</sub>	724.00		NIST Webbook
r <sub>inpol</sub>	726.00		NIST Webbook
r <sub>inpol</sub>	720.00		NIST Webbook
r <sub>inpol</sub>	718.00		NIST Webbook

rinpol	720.00	NIST Webbook
rinpol	704.00	NIST Webbook
rinpol	726.00	NIST Webbook
rinpol	719.00	NIST Webbook
rinpol	718.00	NIST Webbook
rinpol	725.00	NIST Webbook
rinpol	690.20	NIST Webbook
rinpol	723.00	NIST Webbook
rinpol	690.90	NIST Webbook
rinpol	692.00	NIST Webbook
rinpol	722.30	NIST Webbook
rinpol	723.40	NIST Webbook
rinpol	735.00	NIST Webbook
rinpol	722.00	NIST Webbook
rinpol	707.00	NIST Webbook
rinpol	717.00	NIST Webbook
rinpol	688.80	NIST Webbook
rinpol	720.00	NIST Webbook
rinpol	716.00	NIST Webbook
rinpol	720.00	NIST Webbook
rinpol	722.00	NIST Webbook
rinpol	716.00	NIST Webbook
rinpol	718.00	NIST Webbook
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rinpol	717.00	NIST Webbook
rinpol	718.00	NIST Webbook
rinpol	725.00	NIST Webbook
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rinpol	718.00	NIST Webbook
rinpol	718.00	NIST Webbook
rinpol	723.00	NIST Webbook
ripol	1087.00	NIST Webbook
ripol	1101.00	NIST Webbook
ripol	1110.00	NIST Webbook
ripol	1090.00	NIST Webbook
ripol	1094.00	NIST Webbook
ripol	1079.00	NIST Webbook
ripol	1114.00	NIST Webbook
ripol	1112.00	NIST Webbook
ripol	1097.00	NIST Webbook

ripol	1101.00			NIST Webbook
ripol	1101.00			NIST Webbook
tb	395.95 ± 1.00	K		NIST Webbook
tb	395.65 ± 1.00	K		NIST Webbook
tb	393.15 ± 2.00	K		NIST Webbook
tb	394.65 ± 1.00	K		NIST Webbook
tb	394.20	K		NIST Webbook
tb	393.65 ± 2.00	K		NIST Webbook
tb	394.24 ± 0.20	K		NIST Webbook
tb	396.65 ± 4.00	K		NIST Webbook
tb	392.65 ± 4.00	K		NIST Webbook
tb	396.15 ± 2.00	K		NIST Webbook
tb	395.50 ± 1.00	K		NIST Webbook
tb	395.45 ± 2.00	K		NIST Webbook
tb	394.65 ± 2.00	K		NIST Webbook
tb	395.15 ± 1.00	K		NIST Webbook
tb	397.15 ± 1.00	K		NIST Webbook
tb	394.20	K		NIST Webbook
tb	395.15 ± 2.00	K		NIST Webbook
tb	394.65 ± 2.00	K		NIST Webbook
tb	395.00 ± 2.00	K		NIST Webbook
tc	559.50	K		NIST Webbook
tc	559.50 ± 0.70	K		NIST Webbook
tc	559.50 ± 0.70	K		NIST Webbook
tf	170.48	K		Aqueous Solubility Prediction Method
vc	0.380		m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	240.32	J/mol×K	511.13	Joback Method
cpg	230.71	J/mol×K	482.63	Joback Method
cpg	220.61	J/mol×K	454.13	Joback Method
cpg	266.42	J/mol×K	596.64	Joback Method
cpg	258.15	J/mol×K	568.14	Joback Method
cpg	249.46	J/mol×K	539.64	Joback Method
cpg	210.01	J/mol×K	425.63	Joback Method
cpl	289.03	J/mol×K	298.15	NIST Webbook

dvisc	0.0020600	Paxs	313.15	Density and Viscosity Measurement of n-Butylamine with Hexyl Alcohol Isomer Binary Systems
dvisc	0.0014600	Paxs	323.15	Density and Viscosity Measurement of n-Butylamine with Hexyl Alcohol Isomer Binary Systems
dvisc	0.0029300	Paxs	303.15	Density and Viscosity Measurement of n-Butylamine with Hexyl Alcohol Isomer Binary Systems
hvapt	58.30	kJ/mol	342.00	NIST Webbook
hvapt	44.40 ± 0.20	kJ/mol	368.00	NIST Webbook
hvapt	51.30	kJ/mol	342.00	NIST Webbook
hvapt	39.59	kJ/mol	394.20	NIST Webbook
hvapt	44.20	kJ/mol	368.50	NIST Webbook
hvapt	48.90	kJ/mol	363.50	NIST Webbook
hvapt	46.10 ± 0.20	kJ/mol	358.00	NIST Webbook
hvapt	48.50 ± 0.20	kJ/mol	343.00	NIST Webbook
hvapt	50.70 ± 0.20	kJ/mol	328.00	NIST Webbook
hvapt	52.80 ± 0.20	kJ/mol	313.00	NIST Webbook
hvapt	49.10	kJ/mol	331.00	NIST Webbook
rhol	793.50	kg/m3	323.15	Density and Viscosity Measurement of n-Butylamine with Hexyl Alcohol Isomer Binary Systems
rhol	802.40	kg/m3	313.15	Density and Viscosity Measurement of n-Butylamine with Hexyl Alcohol Isomer Binary Systems
rhol	810.80	kg/m3	303.15	Density and Viscosity Measurement of n-Butylamine with Hexyl Alcohol Isomer Binary Systems

# Sources

<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Density and Viscosity Measurement of n-Butylamine with Hexyl Alcohol Joback Method:</b>	<a href="https://www.doi.org/10.1021/je0498053">https://www.doi.org/10.1021/je0498053</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>Aqueous Solubility Prediction Method:</b>	<a href="http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDataset002.xlsx">http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDataset002.xlsx</a>
<b>Estimated Solubility Method:</b>	<a href="http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl_file/ci034243xsi20040112_053635.txt">http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl_file/ci034243xsi20040112_053635.txt</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C590363&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C590363&amp;Units=SI</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>cpl:</b>	Liquid phase heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rho:</b>	Liquid Density
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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