

# Benzene, (isothiocyanatomethyl)-

<b>Other names:</b>	AB 2 Benzyl Isothiocyanate Benzyl mustard oil Benzylsenfoel Isothiocyanic acid, benzyl ester NSC 118976 Toluene, «alpha»-isothiocyanato- Toluene, Å«alphaÅ»-isothiocyanato- Tromacaps Tromalyt Urogran
<b>Inchi:</b>	InChI=1S/C8H7NS/c10-7-9-6-8-4-2-1-3-5-8/h1-5H,6H2
<b>InchiKey:</b>	MDKCFLQDBWCQCV-UHFFFAOYSA-N
<b>Formula:</b>	C8H7NS
<b>SMILES:</b>	S=C=NCc1ccccc1
<b>Mol. weight [g/mol]:</b>	149.21
<b>CAS:</b>	622-78-6

## Physical Properties

Property code	Value	Unit	Source
hf	312.15	kJ/mol	Joback Method
hvap	46.12	kJ/mol	Joback Method
log10ws	-2.70		Crippen Method
logp	2.289		Crippen Method
mcvol	117.550	ml/mol	McGowan Method
pc	3749.97	kPa	Joback Method
rinpol	1354.00		NIST Webbook
rinpol	1335.00		NIST Webbook
rinpol	1318.00		NIST Webbook
rinpol	1334.00		NIST Webbook
rinpol	1353.00		NIST Webbook
rinpol	1377.90		NIST Webbook
rinpol	1317.00		NIST Webbook
rinpol	1361.00		NIST Webbook
rinpol	1377.90		NIST Webbook
rinpol	1354.00		NIST Webbook
rinpol	1391.00		NIST Webbook

ripol	1389.00			NIST Webbook
ripol	1391.00			NIST Webbook
ripol	1317.00			NIST Webbook
ripol	1318.00			NIST Webbook
ripol	1359.00			NIST Webbook
ripol	1389.00			NIST Webbook
ripol	1317.00			NIST Webbook
ripol	2109.00			NIST Webbook
ripol	2107.00			NIST Webbook
ripol	2131.00			NIST Webbook
ripol	2130.00			NIST Webbook
ripol	2130.00			NIST Webbook
ripol	2109.00			NIST Webbook
ripol	2071.00			NIST Webbook
tb	516.00		K	NIST Webbook
tc	814.89		K	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
hvapt	62.20	kJ/mol	434.00	NIST Webbook

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	398.50 ± 0.50	K	1.60	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.81400e+01
Coeff. B	-6.77895e+03
Coeff. C	-1.48100e+01

Temperature range (K), min.	352.00
Temperature range (K), max.	543.24

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C622786&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C622786&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature

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