

# Heptane, 3-chloro-3-methyl-

<b>Other names:</b>	3-Chloro-3-methylheptane
<b>Inchi:</b>	InChI=1S/C8H17Cl/c1-4-6-7-8(3,9)5-2/h4-7H2,1-3H3
<b>InchiKey:</b>	XGAHNVWATDXDPG-UHFFFAOYSA-N
<b>Formula:</b>	C8H17Cl
<b>SMILES:</b>	CCCCC(C)(Cl)CC
<b>Mol. weight [g/mol]:</b>	148.67
<b>CAS:</b>	5272-02-6

## Physical Properties

Property code	Value	Unit	Source
gf	7.39	kJ/mol	Joback Method
hf	-232.94	kJ/mol	Joback Method
hfus	13.26	kJ/mol	Joback Method
hvap	36.49	kJ/mol	Joback Method
log10ws	-3.44		Crippen Method
logp	3.584		Crippen Method
mcvol	135.820	ml/mol	McGowan Method
pc	2490.03	kPa	Joback Method
tb	416.64	K	Joback Method
tc	598.52	K	Joback Method
tf	212.26	K	Joback Method
vc	0.521	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	264.68	J/mol×K	416.64	Joback Method
cpg	328.78	J/mol×K	568.20	Joback Method
cpg	317.26	J/mol×K	537.89	Joback Method
cpg	305.11	J/mol×K	507.58	Joback Method
cpg	292.32	J/mol×K	477.27	Joback Method
cpg	278.85	J/mol×K	446.95	Joback Method
cpg	339.71	J/mol×K	598.52	Joback Method
dvisc	0.0002979	Paxs	416.64	Joback Method

dvisc	0.0004084	Paxs	382.58	Joback Method
dvisc	0.0005954	Paxs	348.51	Joback Method
dvisc	0.0009419	Paxs	314.45	Joback Method
dvisc	0.0016656	Paxs	280.39	Joback Method
dvisc	0.0034485	Paxs	246.32	Joback Method
dvisc	0.0090186	Paxs	212.26	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.39315e+01
Coeff. B	-3.62925e+03
Coeff. C	-6.21220e+01
Temperature range (K), min.	328.12
Temperature range (K), max.	483.15

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C5272026&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C5272026&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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