

# 9,10-Anthracenedione, 2-(1,1-dimethylethyl)-

<b>Other names:</b>	2-(tert-butyl)anthracene-9,10-dione 2-tert-Butyl-9,10-anthraquinone 2-tert-butylanthraquinone Anthraquinone, 2-tert-butyl- «beta»-tert-Butylanthraquinone
<b>Inchi:</b>	InChI=1S/C18H16O2/c1-18(2,3)11-8-9-14-15(10-11)17(20)13-7-5-4-6-12(13)16(14)19/h4
<b>InchiKey:</b>	YTPSFXZMJKMUJE-UHFFFAOYSA-N
<b>Formula:</b>	C18H16O2
<b>SMILES:</b>	CC(C)(C)c1ccc2c(c1)C(=O)c1cccc1C2=O
<b>Mol. weight [g/mol]:</b>	264.32
<b>CAS:</b>	84-47-9

## Physical Properties

Property code	Value	Unit	Source
ea	1.56 ± 0.06	eV	NIST Webbook
gf	134.83	kJ/mol	Joback Method
hf	-161.05	kJ/mol	Joback Method
hfus	20.06	kJ/mol	Joback Method
hvap	69.45	kJ/mol	Joback Method
log10ws	-4.95		Crippen Method
logp	3.760		Crippen Method
mcvol	209.240	ml/mol	McGowan Method
pc	2263.26	kPa	Joback Method
tb	819.09	K	Joback Method
tc	1084.31	K	Joback Method
tf	547.58	K	Joback Method
vc	0.796	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	682.68	J/mol×K	1040.11	Joback Method
cpg	616.33	J/mol×K	819.09	Joback Method
cpg	632.14	J/mol×K	863.29	Joback Method

cpg	646.59	J/mol×K	907.50	Joback Method
cpg	659.76	J/mol×K	951.70	Joback Method
cpg	671.76	J/mol×K	995.90	Joback Method
cpg	692.61	J/mol×K	1084.31	Joback Method
hvapt	101.40	kJ/mol	503.00	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Solubility of 2-Ethylantraquinone in Binary Mixtures of Oligooxymethylene Dimethyl Ethers with Different Number of CH<sub>2</sub>O Groups of n = 2, 3, and 4 from 293.15 to 343.15 K:</b>	<a href="https://www.doi.org/10.1021/acs.jced.6b00334">https://www.doi.org/10.1021/acs.jced.6b00334</a>
<b>Joback Method</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C84479&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C84479&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>ea:</b>	Electron affinity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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