

Dimethylmalonic acid, monochloride, 4-octyl ester

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|----------------------|---|
| Inchi: | InChI=1S/C13H23ClO3/c1-5-7-9-10(8-6-2)17-12(16)13(3,4)11(14)15/h10H,5-9H2,1-4H3 |
| InchiKey: | KOACAVSGVBGTGN-UHFFFAOYSA-N |
| Formula: | C13H23ClO3 |
| SMILES: | CCCCC(CCC)OC(=O)C(C)(C)C(=O)Cl |
| Mol. weight [g/mol]: | 262.77 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|----------------------|----------------|
| gf | -315.79 | kJ/mol | Joback Method |
| hf | -698.80 | kJ/mol | Joback Method |
| hfus | 27.07 | kJ/mol | Joback Method |
| hvap | 63.14 | kJ/mol | Joback Method |
| log10ws | -3.93 | | Crippen Method |
| logp | 3.680 | | Crippen Method |
| mvol | 215.280 | ml/mol | McGowan Method |
| pc | 1784.86 | kPa | Joback Method |
| rinpol | 1477.00 | | NIST Webbook |
| tb | 660.76 | K | Joback Method |
| tc | 852.92 | K | Joback Method |
| tf | 375.70 | K | Joback Method |
| vc | 0.826 | m ³ /kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|-----------|---------|-----------------|---------------|
| cpg | 574.05 | J/mol×K | 660.76 | Joback Method |
| cpg | 642.21 | J/mol×K | 820.90 | Joback Method |
| cpg | 630.17 | J/mol×K | 788.87 | Joback Method |
| cpg | 617.36 | J/mol×K | 756.84 | Joback Method |
| cpg | 603.76 | J/mol×K | 724.81 | Joback Method |
| cpg | 589.33 | J/mol×K | 692.79 | Joback Method |
| cpg | 653.51 | J/mol×K | 852.92 | Joback Method |
| dvisc | 0.0001266 | Paxs | 660.76 | Joback Method |
| dvisc | 0.0001716 | Paxs | 613.25 | Joback Method |

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|-------|-----------|------|--------|---------------|
| dvisc | 0.0002450 | Paxs | 565.74 | Joback Method |
| dvisc | 0.0003733 | Paxs | 518.23 | Joback Method |
| dvisc | 0.0006192 | Paxs | 470.72 | Joback Method |
| dvisc | 0.0011508 | Paxs | 423.21 | Joback Method |
| dvisc | 0.0025016 | Paxs | 375.70 | Joback Method |

Sources

| | |
|------------------------|---|
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=U361742&Units=SI |
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci990307l |
| Crippen Method: | https://www.cheméo.com/doc/models/crippen_log10ws |

Legend

| | |
|-----------------|---|
| cpg: | Ideal gas heat capacity |
| dvisc: | Dynamic viscosity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| rinpol: | Non-polar retention indices |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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