

7-Oxabicyclo[2.2.1]heptane-2,5-dione, 3,6-dimethylidene-

Inchi:	InChI=1S/C8H6O3/c1-3-5(9)8-4(2)6(10)7(3)11-8/h7-8H,1-2H2
InchiKey:	YIHLCAOATGPYBG-UHFFFAOYSA-N
Formula:	C8H6O3
SMILES:	C=C1C(=O)C2OC1C(=O)C2=C
Mol. weight [g/mol]:	150.13
CAS:	127750-98-5

Physical Properties

Property code	Value	Unit	Source
gf	-99.26	kJ/mol	Joback Method
hf	-307.93	kJ/mol	Joback Method
hfus	15.33	kJ/mol	Joback Method
hvap	46.72	kJ/mol	Joback Method
ie	9.40	eV	NIST Webbook
ie	9.10	eV	NIST Webbook
log10ws	-0.54		Crippen Method
logp	0.018		Crippen Method
mcvol	102.270	ml/mol	McGowan Method
pc	3896.50	kPa	Joback Method
tb	561.10	K	Joback Method
tc	800.43	K	Joback Method
tf	402.65	K	Joback Method
vc	0.393	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	250.41	J/mol×K	561.10	Joback Method
cpg	262.49	J/mol×K	600.99	Joback Method
cpg	273.95	J/mol×K	640.88	Joback Method
cpg	284.80	J/mol×K	680.76	Joback Method
cpg	295.00	J/mol×K	720.65	Joback Method
cpg	304.55	J/mol×K	760.54	Joback Method
cpg	313.43	J/mol×K	800.43	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C127750985&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
ie:	Ionization energy
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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