

# (pentyloxy)benzene

<b>Inchi:</b>	InChI=1S/C11H16O/c1-2-3-7-10-12-11-8-5-4-6-9-11/h4-6,8-9H,2-3,7,10H2,1H3
<b>InchiKey:</b>	HPUOAJPGWQQRNT-UHFFFAOYSA-N
<b>Formula:</b>	C11H16O
<b>SMILES:</b>	CCCCCOc1ccccc1
<b>Mol. weight [g/mol]:</b>	164.24
<b>CAS:</b>	2050-04-6

## Physical Properties

Property code	Value	Unit	Source
gf	49.15	kJ/mol	Joback Method
hf	-166.06	kJ/mol	Joback Method
hfus	19.47	kJ/mol	Joback Method
hvap	44.77	kJ/mol	Joback Method
log10ws	-3.27		Crippen Method
logp	3.256		Crippen Method
mcvol	147.960	ml/mol	McGowan Method
pc	2603.08	kPa	Joback Method
tb	500.18	K	Joback Method
tc	700.57	K	Joback Method
tf	262.38	K	Joback Method
vc	0.561	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	325.53	J/molxK	500.18	Joback Method
cpg	341.15	J/molxK	533.58	Joback Method
cpg	355.99	J/molxK	566.98	Joback Method
cpg	370.07	J/molxK	600.38	Joback Method
cpg	383.41	J/molxK	633.78	Joback Method
cpg	396.02	J/molxK	667.17	Joback Method
cpg	407.94	J/molxK	700.57	Joback Method
dvisc	0.0028054	Paxs	262.38	Joback Method
dvisc	0.0013237	Paxs	302.01	Joback Method

dvisc	0.0007435	Paxs	341.65	Joback Method
dvisc	0.0004708	Paxs	381.28	Joback Method
dvisc	0.0003249	Paxs	420.91	Joback Method
dvisc	0.0002390	Paxs	460.55	Joback Method
dvisc	0.0001846	Paxs	500.18	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2050046&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2050046&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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