

# 1,1,3-trichlorobutane

|                      |                                               |
|----------------------|-----------------------------------------------|
| Inchi:               | InChI=1S/C4H7Cl3/c1-3(5)2-4(6)7/h3-4H,2H2,1H3 |
| InchiKey:            | GSBUVYHAANRCNO-UHFFFAOYSA-N                   |
| Formula:             | C4H7Cl3                                       |
| SMILES:              | CC(Cl)CC(Cl)Cl                                |
| Mol. weight [g/mol]: | 161.46                                        |
| CAS:                 | 13279-87-3                                    |

## Physical Properties

| Property code | Value         | Unit    | Source         |
|---------------|---------------|---------|----------------|
| gf            | -57.87        | kJ/mol  | Joback Method  |
| hf            | -183.67       | kJ/mol  | Joback Method  |
| hfus          | 11.66         | kJ/mol  | Joback Method  |
| hvap          | 36.88         | kJ/mol  | Joback Method  |
| log10ws       | -2.68         |         | Crippen Method |
| logp          | 2.807         |         | Crippen Method |
| mcvol         | 103.940       | ml/mol  | McGowan Method |
| pc            | 3472.45       | kPa     | Joback Method  |
| tb            | 423.00 ± 4.00 | K       | NIST Webbook   |
| tb            | 423.00 ± 3.00 | K       | NIST Webbook   |
| tc            | 601.91        | K       | Joback Method  |
| tf            | 194.60        | K       | Joback Method  |
| vc            | 0.395         | m3/kmol | Joback Method  |

## Temperature Dependent Properties

| Property code | Value     | Unit    | Temperature [K] | Source        |
|---------------|-----------|---------|-----------------|---------------|
| cpg           | 163.69    | J/mol×K | 402.33          | Joback Method |
| cpg           | 198.08    | J/mol×K | 568.64          | Joback Method |
| cpg           | 191.94    | J/mol×K | 535.38          | Joback Method |
| cpg           | 185.45    | J/mol×K | 502.12          | Joback Method |
| cpg           | 178.58    | J/mol×K | 468.86          | Joback Method |
| cpg           | 171.34    | J/mol×K | 435.59          | Joback Method |
| cpg           | 203.87    | J/mol×K | 601.91          | Joback Method |
| dvisc         | 0.0003596 | Pa×s    | 402.33          | Joback Method |

|       |           |      |        |               |
|-------|-----------|------|--------|---------------|
| dvisc | 0.0004848 | Pa×s | 367.71 | Joback Method |
| dvisc | 0.0006955 | Pa×s | 333.09 | Joback Method |
| dvisc | 0.0010849 | Pa×s | 298.47 | Joback Method |
| dvisc | 0.0019018 | Pa×s | 263.84 | Joback Method |
| dvisc | 0.0039497 | Pa×s | 229.22 | Joback Method |
| dvisc | 0.0106392 | Pa×s | 194.60 | Joback Method |

## Sources

|                        |                                                                                                                                               |
|------------------------|-----------------------------------------------------------------------------------------------------------------------------------------------|
| <b>Crippen Method:</b> | <a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>                                     |
| <b>Crippen Method:</b> | <a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>                             |
| <b>Joback Method:</b>  | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>                                         |
| <b>McGowan Method:</b> | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>                         |
| <b>NIST Webbook:</b>   | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C13279873&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C13279873&amp;Units=SI</a> |

## Legend

|                 |                                                 |
|-----------------|-------------------------------------------------|
| <b>cpg:</b>     | Ideal gas heat capacity                         |
| <b>dvisc:</b>   | Dynamic viscosity                               |
| <b>gf:</b>      | Standard Gibbs free energy of formation         |
| <b>hf:</b>      | Enthalpy of formation at standard conditions    |
| <b>hfus:</b>    | Enthalpy of fusion at standard conditions       |
| <b>hvap:</b>    | Enthalpy of vaporization at standard conditions |
| <b>log10ws:</b> | Log10 of Water solubility in mol/l              |
| <b>logp:</b>    | Octanol/Water partition coefficient             |
| <b>mccvol:</b>  | McGowan's characteristic volume                 |
| <b>pc:</b>      | Critical Pressure                               |
| <b>tb:</b>      | Normal Boiling Point Temperature                |
| <b>tc:</b>      | Critical Temperature                            |
| <b>tf:</b>      | Normal melting (fusion) point                   |
| <b>vc:</b>      | Critical Volume                                 |

Latest version available from:

<https://www.chemeo.com/cid/110-779-5/1-1-3-trichlorobutane.pdf>

Generated by Cheméo on 2025-12-05 17:20:29.546990112 +0000 UTC m=+4703427.077030772.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.