

Propanamide, N-(2,5-dimethoxyphenyl)-3-phenyl-

Inchi:	InChI=1S/C17H19NO3/c1-20-14-9-10-16(21-2)15(12-14)18-17(19)11-8-13-6-4-3-5-7-13/
InchiKey:	LFAIOQIJRNRAIK-UHFFFAOYSA-N
Formula:	C17H19NO3
SMILES:	COc1ccc(OC)c(NC(=O)CCc2ccccc2)c1
Mol. weight [g/mol]:	285.34

Physical Properties

Property code	Value	Unit	Source
gf	48.29	kJ/mol	Joback Method
hf	-267.64	kJ/mol	Joback Method
hfus	36.16	kJ/mol	Joback Method
hvap	77.31	kJ/mol	Joback Method
log10ws	-3.95		Crippen Method
logp	3.275		Crippen Method
mcvol	226.160	ml/mol	McGowan Method
pc	2111.94	kPa	Joback Method
rinpol	2454.00		NIST Webbook
rinpol	2454.00		NIST Webbook
tb	800.56	K	Joback Method
tc	1026.74	K	Joback Method
tf	506.28	K	Joback Method
vc	0.849	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	651.94	J/mol×K	800.56	Joback Method
cpg	666.61	J/mol×K	838.26	Joback Method
cpg	680.07	J/mol×K	875.95	Joback Method
cpg	692.34	J/mol×K	913.65	Joback Method
cpg	703.46	J/mol×K	951.35	Joback Method
cpg	713.46	J/mol×K	989.04	Joback Method
cpg	722.35	J/mol×K	1026.74	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U308119&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinp:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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