

Benzamide, 2,4,5-trifluoro-3-methoxy-N-octyl-

Inchi:	InChI=1S/C16H22F3NO2/c1-3-4-5-6-7-8-9-20-16(21)11-10-12(17)14(19)15(22-2)13(11)1
InchiKey:	BIIYMFJZPDYYMU-UHFFFAOYSA-N
Formula:	C16H22F3NO2
SMILES:	CCCCCCCCNC(=O)c1cc(F)c(F)c(OC)c1F
Mol. weight [g/mol]:	317.35

Physical Properties

Property code	Value	Unit	Source
gf	-571.23	kJ/mol	Joback Method
hf	-962.58	kJ/mol	Joback Method
hfus	46.81	kJ/mol	Joback Method
hvap	69.28	kJ/mol	Joback Method
log10ws	-5.86		Crippen Method
logp	4.203		Crippen Method
mcvol	235.270	ml/mol	McGowan Method
pc	1527.07	kPa	Joback Method
rinpol	2141.00		NIST Webbook
rinpol	2141.00		NIST Webbook
tb	736.35	K	Joback Method
tc	918.99	K	Joback Method
tf	473.17	K	Joback Method
vc	0.936	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	679.42	J/mol×K	736.35	Joback Method
cpg	693.89	J/mol×K	766.79	Joback Method
cpg	707.57	J/mol×K	797.23	Joback Method
cpg	720.49	J/mol×K	827.67	Joback Method
cpg	732.65	J/mol×K	858.11	Joback Method
cpg	744.07	J/mol×K	888.55	Joback Method
cpg	754.76	J/mol×K	918.99	Joback Method

Sources

Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U407645&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpola:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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