

Sarcosine, N-(4-methylbenzoyl)-, isobutyl ester

Inchi:	InChI=1S/C15H21NO3/c1-11(2)10-19-14(17)9-16(4)15(18)13-7-5-12(3)6-8-13/h5-8,11H,
InchiKey:	IHAMTAGWELIWOU-UHFFFAOYSA-N
Formula:	C15H21NO3
SMILES:	<chem>Cc1ccc(C(=O)N(C)CC(=O)OCC(C)C)cc1</chem>
Mol. weight [g/mol]:	263.33

Physical Properties

Property code	Value	Unit	Source
gf	-76.30	kJ/mol	Joback Method
hf	-423.00	kJ/mol	Joback Method
hfus	32.14	kJ/mol	Joback Method
hvap	69.48	kJ/mol	Joback Method
log10ws	-2.80		Crippen Method
logp	2.266		Crippen Method
mvol	217.440	ml/mol	McGowan Method
pc	2016.32	kPa	Joback Method
rinpol	2066.00		NIST Webbook
rinpol	2066.00		NIST Webbook
tb	716.42	K	Joback Method
tc	922.64	K	Joback Method
tf	437.31	K	Joback Method
vc	0.809	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	606.64	J/mol×K	716.42	Joback Method
cpg	622.19	J/mol×K	750.79	Joback Method
cpg	636.72	J/mol×K	785.16	Joback Method
cpg	650.28	J/mol×K	819.53	Joback Method
cpg	662.89	J/mol×K	853.90	Joback Method
cpg	674.60	J/mol×K	888.27	Joback Method
cpg	685.42	J/mol×K	922.64	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U321214&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
h vap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
r in pol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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