

Oxetane, 3-bromomethyl-3-(2-chlorophenoxy)methyl

Inchi:	InChI=1S/C11H12BrClO2/c12-5-11(6-14-7-11)8-15-10-4-2-1-3-9(10)13/h1-4H,5-8H2
InchiKey:	GJBLFJPTYOCVOG-UHFFFAOYSA-N
Formula:	C11H12BrClO2
SMILES:	Clc1ccccc1OCC1(CBr)COC1
Mol. weight [g/mol]:	291.57

Physical Properties

Property code	Value	Unit	Source
gf	-1.05	kJ/mol	Joback Method
hf	-217.06	kJ/mol	Joback Method
hfus	26.28	kJ/mol	Joback Method
hvap	59.69	kJ/mol	Joback Method
log10ws	-3.12		Crippen Method
logp	3.130		Crippen Method
mcvol	172.710	ml/mol	McGowan Method
pc	3257.86	kPa	Joback Method
rinpol	1884.00		NIST Webbook
rinpol	1884.00		NIST Webbook
tb	646.95	K	Joback Method
tc	896.66	K	Joback Method
tf	429.51	K	Joback Method
vc	0.640	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	406.59	J/mol×K	646.95	Joback Method
cpg	420.28	J/mol×K	688.57	Joback Method
cpg	433.12	J/mol×K	730.19	Joback Method
cpg	445.31	J/mol×K	771.81	Joback Method
cpg	457.04	J/mol×K	813.42	Joback Method
cpg	468.52	J/mol×K	855.04	Joback Method
cpg	479.96	J/mol×K	896.66	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R6609&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinp:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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