

Glutaric acid, 2-bromo-5-fluorobenzyl propyl ester

Inchi:	InChI=1S/C15H18BrFO4/c1-2-8-20-14(18)4-3-5-15(19)21-10-11-9-12(17)6-7-13(11)16/h
InchiKey:	LVPUBKGDSEYPGY-UHFFFAOYSA-N
Formula:	C15H18BrFO4
SMILES:	CCCOC(=O)CCCC(=O)OCc1cc(F)ccc1Br
Mol. weight [g/mol]:	361.20

Physical Properties

Property code	Value	Unit	Source
gf	-479.76	kJ/mol	Joback Method
hf	-798.72	kJ/mol	Joback Method
hfus	41.81	kJ/mol	Joback Method
hvap	76.51	kJ/mol	Joback Method
log10ws	-4.92		Crippen Method
logp	3.755		Crippen Method
mcvol	232.600	ml/mol	McGowan Method
pc	1991.21	kPa	Joback Method
rinpola	2211.00		NIST Webbook
rinpola	2211.00		NIST Webbook
tb	797.25	K	Joback Method
tc	1005.05	K	Joback Method
tf	514.98	K	Joback Method
vc	0.895	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	638.56	J/molxK	797.25	Joback Method
cpg	651.18	J/molxK	831.88	Joback Method
cpg	662.90	J/molxK	866.52	Joback Method
cpg	673.72	J/molxK	901.15	Joback Method
cpg	683.65	J/molxK	935.78	Joback Method
cpg	692.71	J/molxK	970.42	Joback Method
cpg	700.91	J/molxK	1005.05	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U377059&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinp:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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