

# Glutaric acid, 2,3-dichlorophenyl 3-octyl ester

<b>Inchi:</b>	InChI=1S/C19H26Cl2O4/c1-3-5-6-9-14(4-2)24-17(22)12-8-13-18(23)25-16-11-7-10-15(20)
<b>InchiKey:</b>	GBLSADZZJBTMAT-UHFFFAOYSA-N
<b>Formula:</b>	C19H26Cl2O4
<b>SMILES:</b>	CCCCC(CC)OC(=O)CCCC(=O)Oc1cccc(Cl)c1Cl
<b>Mol. weight [g/mol]:</b>	389.31

## Physical Properties

Property code	Value	Unit	Source
gf	-291.89	kJ/mol	Joback Method
hf	-748.26	kJ/mol	Joback Method
hfus	48.67	kJ/mol	Joback Method
hvap	88.18	kJ/mol	Joback Method
log10ws	-6.73		Crippen Method
logp	5.971		Crippen Method
mvol	294.170	ml/mol	McGowan Method
pc	1356.63	kPa	Joback Method
rinpol	2582.00		NIST Webbook
rinpol	2582.00		NIST Webbook
tb	897.76	K	Joback Method
tc	1109.78	K	Joback Method
tf	544.51	K	Joback Method
vc	1.131	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	870.96	J/molxK	897.76	Joback Method
cpg	927.40	J/molxK	1074.44	Joback Method
cpg	918.36	J/molxK	1039.10	Joback Method
cpg	908.22	J/molxK	1003.77	Joback Method
cpg	896.95	J/molxK	968.43	Joback Method
cpg	884.54	J/molxK	933.10	Joback Method
cpg	935.37	J/molxK	1109.78	Joback Method
dvisc	0.0000434	Paxs	897.76	Joback Method

dvisc	0.0000555	Paxs	838.88	Joback Method
dvisc	0.0000737	Paxs	780.01	Joback Method
dvisc	0.0001025	Paxs	721.13	Joback Method
dvisc	0.0001510	Paxs	662.26	Joback Method
dvisc	0.0002401	Paxs	603.38	Joback Method
dvisc	0.0004221	Paxs	544.51	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U391568&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U391568&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/113-092-4/Glutaric-acid-2-3-dichlorophenyl-3-octyl-ester.pdf>

Generated by Cheméo on 2024-04-27 22:03:43.607104357 +0000 UTC m=+16544672.527681672.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.