

Succinic acid, 3,4-difluorobenzyl pentyl ester

Inchi:	InChI=1S/C16H20F2O4/c1-2-3-4-9-21-15(19)7-8-16(20)22-11-12-5-6-13(17)14(18)10-12
InchiKey:	JQFKHHSAPYXPA-UHFFFAOYSA-N
Formula:	C16H20F2O4
SMILES:	CCCCCOC(=O)CCC(=O)OCc1ccc(F)c(F)c1
Mol. weight [g/mol]:	314.32

Physical Properties

Property code	Value	Unit	Source
gf	-680.47	kJ/mol	Joback Method
hf	-1041.80	kJ/mol	Joback Method
hfus	42.19	kJ/mol	Joback Method
hvap	71.49	kJ/mol	Joback Method
log10ws	-4.51		Crippen Method
logp	3.522		Crippen Method
mvol	230.960	ml/mol	McGowan Method
pc	1664.61	kPa	Joback Method
rinpol	2024.00		NIST Webbook
rinpol	2024.00		NIST Webbook
tb	753.24	K	Joback Method
tc	943.80	K	Joback Method
tf	467.04	K	Joback Method
vc	0.907	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	663.08	J/molxK	753.24	Joback Method
cpg	677.05	J/molxK	785.00	Joback Method
cpg	690.16	J/molxK	816.76	Joback Method
cpg	702.41	J/molxK	848.52	Joback Method
cpg	713.82	J/molxK	880.28	Joback Method
cpg	724.40	J/molxK	912.04	Joback Method
cpg	734.14	J/molxK	943.80	Joback Method

Sources

NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U381743&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinp:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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