

# 2,2-dichloroethyl hexadecanoate

**Inchi:** InChI=1S/C18H34Cl2O2/c1-2-3-4-5-6-7-8-9-10-11-12-13-14-15-18(21)22-16-17(19)20/h  
**InchiKey:** KHCHNMQCCZLLSQF-UHFFFAOYSA-N  
**Formula:** C18H34Cl2O2  
**SMILES:** CCCCCCCCCCCCCCCC(=O)OCC(Cl)Cl  
**Mol. weight [g/mol]:** 353.37

## Physical Properties

Property code	Value	Unit	Source
gf	-159.54	kJ/mol	Joback Method
hf	-696.41	kJ/mol	Joback Method
hfus	50.03	kJ/mol	Joback Method
hvap	73.20	kJ/mol	Joback Method
log10ws	-7.13		Crippen Method
logp	6.815		Crippen Method
mcvol	296.400	ml/mol	McGowan Method
pc	1136.73	kPa	Joback Method
ripol	2301.00		NIST Webbook
ripol	2301.00		NIST Webbook
ripol	2301.00		NIST Webbook
ripol	2293.00		NIST Webbook
ripol	2293.00		NIST Webbook
ripol	2780.00		NIST Webbook
ripol	2780.00		NIST Webbook
ripol	2765.00		NIST Webbook
ripol	2792.00		NIST Webbook
ripol	2785.00		NIST Webbook
ripol	2781.00		NIST Webbook
ripol	2765.00		NIST Webbook
tb	761.95	K	Joback Method
tc	942.76	K	Joback Method
tf	409.62	K	Joback Method
vc	1.159	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	859.08	J/molxK	761.95	Joback Method
cpg	876.39	J/molxK	792.08	Joback Method
cpg	892.79	J/molxK	822.22	Joback Method
cpg	908.32	J/molxK	852.35	Joback Method
cpg	922.99	J/molxK	882.49	Joback Method
cpg	936.84	J/molxK	912.62	Joback Method
cpg	949.89	J/molxK	942.76	Joback Method
dvisc	0.0015155	Paxs	409.62	Joback Method
dvisc	0.0006523	Paxs	468.34	Joback Method
dvisc	0.0003388	Paxs	527.06	Joback Method
dvisc	0.0002006	Paxs	585.79	Joback Method
dvisc	0.0001307	Paxs	644.51	Joback Method
dvisc	0.0000915	Paxs	703.23	Joback Method
dvisc	0.0000677	Paxs	761.95	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R30634&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R30634&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient

<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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