

# Succinic acid, isohexyl 2,3,4,6-tetrachlorophenyl ester

<b>Inchi:</b>	InChI=1S/C16H18Cl4O4/c1-9(2)4-3-7-23-12(21)5-6-13(22)24-16-11(18)8-10(17)14(19)15
<b>InchiKey:</b>	OKWVSPPIFLQOU-UHFFFAOYSA-N
<b>Formula:</b>	C16H18Cl4O4
<b>SMILES:</b>	CC(C)CCCOC(=O)CCC(=O)Oc1c(Cl)cc(Cl)c(Cl)c1Cl
<b>Mol. weight [g/mol]:</b>	416.12

## Physical Properties

Property code	Value	Unit	Source
gf	-360.27	kJ/mol	Joback Method
hf	-740.76	kJ/mol	Joback Method
hfus	48.52	kJ/mol	Joback Method
hvap	91.60	kJ/mol	Joback Method
log10ws	-6.50		Crippen Method
logp	5.965		Crippen Method
mcvol	276.380	ml/mol	McGowan Method
pc	1561.05	kPa	Joback Method
rinpol	2626.00		NIST Webbook
rinpol	2626.00		NIST Webbook
tb	913.94	K	Joback Method
tc	1136.40	K	Joback Method
tf	595.58	K	Joback Method
vc	1.062	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	740.33	J/molxK	913.94	Joback Method
cpg	750.72	J/molxK	951.02	Joback Method
cpg	760.00	J/molxK	988.09	Joback Method
cpg	768.20	J/molxK	1025.17	Joback Method
cpg	775.31	J/molxK	1062.25	Joback Method
cpg	781.34	J/molxK	1099.33	Joback Method
cpg	786.29	J/molxK	1136.40	Joback Method
dvisc	0.0003117	Paxs	595.58	Joback Method

dvisc	0.0002035	Paxs	648.64	Joback Method
dvisc	0.0001417	Paxs	701.70	Joback Method
dvisc	0.0001038	Paxs	754.76	Joback Method
dvisc	0.0000792	Paxs	807.82	Joback Method
dvisc	0.0000625	Paxs	860.88	Joback Method
dvisc	0.0000507	Paxs	913.94	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U349677&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U349677&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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