

Bromo STP

Inchi:	InChI=1S/C11H16BrNO2/c1-7-9(14-2)6-8(4-5-13)11(15-3)10(7)12/h6H,4-5,13H2,1-3H3
InchiKey:	CFNPHYJWDYQUCU-UHFFFAOYSA-N
Formula:	C11H16BrNO2
SMILES:	COc1cc(CCN)c(OC)c(Br)c1C
Mol. weight [g/mol]:	274.15

Physical Properties

Property code	Value	Unit	Source
gf	-13.60	kJ/mol	Joback Method
hf	-284.04	kJ/mol	Joback Method
hfus	29.59	kJ/mol	Joback Method
hvap	66.90	kJ/mol	Joback Method
log10ws	-3.59		Crippen Method
logp	2.276		Crippen Method
mcvol	181.310	ml/mol	McGowan Method
pc	2738.28	kPa	Joback Method
rinpol	2195.00		NIST Webbook
rinpol	2195.00		NIST Webbook
rinpol	2195.00		NIST Webbook
tb	681.21	K	Joback Method
tc	904.00	K	Joback Method
tf	477.75	K	Joback Method
vc	0.670	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	449.66	J/molxK	681.21	Joback Method
cpg	462.85	J/molxK	718.34	Joback Method
cpg	475.30	J/molxK	755.47	Joback Method
cpg	486.99	J/molxK	792.60	Joback Method
cpg	497.93	J/molxK	829.73	Joback Method
cpg	508.10	J/molxK	866.87	Joback Method
cpg	517.50	J/molxK	904.00	Joback Method

Sources

Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R17234&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307I
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
h vap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
r in pol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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