

# Succinic acid, 2,2-dichloroethyl dec-4-en-1-yl ester

Inchi:	InChI=1S/C16H26Cl2O4/c1-2-3-4-5-6-7-8-9-12-21-15(19)10-11-16(20)22-13-14(17)18/h
InchiKey:	OWWFYFLISFJSOA-VOTSOKGWSA-N
Formula:	C16H26Cl2O4
SMILES:	CCCCC=CCCCOC(=O)CCC(=O)OCC(Cl)Cl
Mol. weight [g/mol]:	353.28

## Physical Properties

Property code	Value	Unit	Source
gf	-330.08	kJ/mol	Joback Method
hf	-782.71	kJ/mol	Joback Method
hfus	47.84	kJ/mol	Joback Method
hvap	77.86	kJ/mol	Joback Method
log10ws	-5.01		Crippen Method
logp	4.573		Crippen Method
mvol	271.360	ml/mol	McGowan Method
pc	1396.46	kPa	Joback Method
rinpol	2301.00		NIST Webbook
rinpol	2301.00		NIST Webbook
tb	796.64	K	Joback Method
tc	989.07	K	Joback Method
tf	454.16	K	Joback Method
vc	1.052	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	769.17	J/molxK	796.64	Joback Method
cpg	783.59	J/molxK	828.71	Joback Method
cpg	797.13	J/molxK	860.78	Joback Method
cpg	809.82	J/molxK	892.86	Joback Method
cpg	821.67	J/molxK	924.93	Joback Method
cpg	832.73	J/molxK	957.00	Joback Method
cpg	843.00	J/molxK	989.07	Joback Method
dvisc	0.0008825	Paxs	454.16	Joback Method

dvisc	0.0004315	Paxs	511.24	Joback Method
dvisc	0.0002436	Paxs	568.32	Joback Method
dvisc	0.0001527	Paxs	625.40	Joback Method
dvisc	0.0001034	Paxs	682.48	Joback Method
dvisc	0.0000744	Paxs	739.56	Joback Method
dvisc	0.0000561	Paxs	796.64	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U391172&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U391172&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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