

# Propanamide, N-(3-nitrophenyl)-2-methyl-

<b>Inchi:</b>	InChI=1S/C10H12N2O3/c1-7(2)10(13)11-8-4-3-5-9(6-8)12(14)15/h3-7H,1-2H3,(H,11,13)
<b>InchiKey:</b>	YARNVNLKNTXUBT-UHFFFAOYSA-N
<b>Formula:</b>	C10H12N2O3
<b>SMILES:</b>	CC(C)C(=O)Nc1cccc([N+](=O)[O-])c1
<b>Mol. weight [g/mol]:</b>	208.21

## Physical Properties

Property code	Value	Unit	Source
gf	129.68	kJ/mol	Joback Method
hf	-99.82	kJ/mol	Joback Method
hfus	29.84	kJ/mol	Joback Method
hvap	70.18	kJ/mol	Joback Method
log10ws	-2.92		Crippen Method
logp	2.189		Crippen Method
mcvol	156.970	ml/mol	McGowan Method
pc	3206.41	kPa	Joback Method
rinpola	1899.00		NIST Webbook
rinpola	1899.00		NIST Webbook
tb	715.30	K	Joback Method
tc	958.30	K	Joback Method
tf	472.60	K	Joback Method
vc	0.605	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	419.60	J/molxK	715.30	Joback Method
cpg	431.83	J/molxK	755.80	Joback Method
cpg	443.07	J/molxK	796.30	Joback Method
cpg	453.37	J/molxK	836.80	Joback Method
cpg	462.77	J/molxK	877.30	Joback Method
cpg	471.34	J/molxK	917.80	Joback Method
cpg	479.11	J/molxK	958.30	Joback Method

# Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U307325&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U307325&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h vap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>r in pol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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