

# Fumaric acid, octyl 2,2,2-trichloroethyl ester

**Inchi:** InChI=1S/C14H21Cl3O4/c1-2-3-4-5-6-7-10-20-12(18)8-9-13(19)21-11-14(15,16)17/h8-9H  
**InchiKey:** ZKKSUBBLUEFJBSP-CMDGGGOBGSA-N  
**Formula:** C14H21Cl3O4  
**SMILES:** CCCCCCOC(=O)C=CC(=O)OCC(Cl)(Cl)Cl  
**Mol. weight [g/mol]:** 359.67

## Physical Properties

Property code	Value	Unit	Source
gf	-353.57	kJ/mol	Joback Method
hf	-760.64	kJ/mol	Joback Method
hfus	42.97	kJ/mol	Joback Method
hvap	76.89	kJ/mol	Joback Method
log10ws	-4.82		Crippen Method
logp	4.360		Crippen Method
mvol	255.420	ml/mol	McGowan Method
pc	1587.28	kPa	Joback Method
rinpol	2172.00		NIST Webbook
rinpol	2172.00		NIST Webbook
tb	785.52	K	Joback Method
tc	986.60	K	Joback Method
tf	478.96	K	Joback Method
vc	0.984	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	684.59	J/molxK	785.52	Joback Method
cpg	740.03	J/molxK	953.09	Joback Method
cpg	730.43	J/molxK	919.58	Joback Method
cpg	720.13	J/molxK	886.06	Joback Method
cpg	709.08	J/molxK	852.55	Joback Method
cpg	697.25	J/molxK	819.03	Joback Method
cpg	748.95	J/molxK	986.60	Joback Method
dvisc	0.0000567	Paxs	785.52	Joback Method

dvisc	0.0000746	Paxs	734.43	Joback Method
dvisc	0.0001022	Paxs	683.33	Joback Method
dvisc	0.0001475	Paxs	632.24	Joback Method
dvisc	0.0002268	Paxs	581.15	Joback Method
dvisc	0.0003790	Paxs	530.05	Joback Method
dvisc	0.0007068	Paxs	478.96	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U348505&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U348505&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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