

Benzamide, 4-nitro-N-ethyl-N-hexyl-

Inchi:	InChI=1S/C15H22N2O3/c1-3-5-6-7-12-16(4-2)15(18)13-8-10-14(11-9-13)17(19)20/h8-11
InchiKey:	SPSZXGFYTTQQXLX-UHFFFAOYSA-N
Formula:	C15H22N2O3
SMILES:	CCCCCN(CC)C(=O)c1ccc([N+](=O)[O-])cc1
Mol. weight [g/mol]:	278.35

Physical Properties

Property code	Value	Unit	Source
gf	195.61	kJ/mol	Joback Method
hf	-183.68	kJ/mol	Joback Method
hfus	44.24	kJ/mol	Joback Method
hvap	77.30	kJ/mol	Joback Method
log10ws	-4.78		Crippen Method
logp	3.637		Crippen Method
mvol	227.420	ml/mol	McGowan Method
pc	1942.37	kPa	Joback Method
rinpol	2389.00		NIST Webbook
rinpol	2389.00		NIST Webbook
tb	792.41	K	Joback Method
tc	1010.62	K	Joback Method
tf	523.76	K	Joback Method
vc	0.874	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	673.48	J/mol×K	792.41	Joback Method
cpg	688.25	J/mol×K	828.78	Joback Method
cpg	701.98	J/mol×K	865.15	Joback Method
cpg	714.74	J/mol×K	901.52	Joback Method
cpg	726.59	J/mol×K	937.88	Joback Method
cpg	737.60	J/mol×K	974.25	Joback Method
cpg	747.81	J/mol×K	1010.62	Joback Method

Sources

Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U415291&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinp:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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