

# Pentafluorobenzoic acid, benzyl ester

<b>Other names:</b>	Benzyl 2,3,4,5,6-pentafluorobenzoate Benzoic acid, pentafluoro-, phenylmethyl ester Benzenemethanol, pentafluorobenzoate
<b>Inchi:</b>	InChI=1S/C14H7F5O2/c15-9-8(10(16)12(18)13(19)11(9)17)14(20)21-6-7-4-2-1-3-5-7/h1-
<b>InchiKey:</b>	WLPUFAZAMBXLPF-UHFFFAOYSA-N
<b>Formula:</b>	C14H7F5O2
<b>SMILES:</b>	O=C(OCc1ccccc1)c1c(F)c(F)c(F)c(F)c1F
<b>Mol. weight [g/mol]:</b>	302.20
<b>CAS:</b>	104856-46-4

## Physical Properties

Property code	Value	Unit	Source
gf	-964.30	kJ/mol	Joback Method
hf	-1141.93	kJ/mol	Joback Method
hfus	36.34	kJ/mol	Joback Method
hvap	59.69	kJ/mol	Joback Method
log10ws	-5.48		Crippen Method
logp	3.739		Crippen Method
mcvol	176.890	ml/mol	McGowan Method
pc	2155.30	kPa	Joback Method
rinpol	1583.00		NIST Webbook
rinpol	1613.00		NIST Webbook
rinpol	1613.00		NIST Webbook
rinpol	1583.00		NIST Webbook
ripol	2180.00		NIST Webbook
ripol	2180.00		NIST Webbook
tb	670.62	K	Joback Method
tc	868.93	K	Joback Method
tf	438.09	K	Joback Method
vc	0.718	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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cpg	450.93	J/mol×K	670.62	Joback Method
cpg	462.10	J/mol×K	703.67	Joback Method
cpg	472.57	J/mol×K	736.72	Joback Method
cpg	482.36	J/mol×K	769.77	Joback Method
cpg	491.46	J/mol×K	802.83	Joback Method
cpg	499.90	J/mol×K	835.88	Joback Method
cpg	507.68	J/mol×K	868.93	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C104856464&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C104856464&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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