

# Dichloroacetic acid, 3,5-dimethylphenyl ester

<b>Inchi:</b>	InChI=1S/C10H10Cl2O2/c1-6-3-7(2)5-8(4-6)14-10(13)9(11)12/h3-5,9H,1-2H3
<b>InchiKey:</b>	NGJFMOALFGIQLY-UHFFFAOYSA-N
<b>Formula:</b>	C10H10Cl2O2
<b>SMILES:</b>	Cc1cc(C)cc(OC(=O)C(Cl)Cl)c1
<b>Mol. weight [g/mol]:</b>	233.09

## Physical Properties

Property code	Value	Unit	Source
gf	-133.75	kJ/mol	Joback Method
hf	-317.70	kJ/mol	Joback Method
hfus	22.58	kJ/mol	Joback Method
hvap	58.99	kJ/mol	Joback Method
log10ws	-3.65		Crippen Method
logp	3.013		Crippen Method
mcvol	159.920	ml/mol	McGowan Method
pc	2778.85	kPa	Joback Method
rinpol	1511.00		NIST Webbook
rinpol	1511.00		NIST Webbook
tb	615.55	K	Joback Method
tc	843.20	K	Joback Method
tf	370.92	K	Joback Method
vc	0.604	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	349.65	J/mol×K	615.55	Joback Method
cpg	400.81	J/mol×K	805.26	Joback Method
cpg	392.02	J/mol×K	767.32	Joback Method
cpg	382.52	J/mol×K	729.37	Joback Method
cpg	372.30	J/mol×K	691.43	Joback Method
cpg	361.35	J/mol×K	653.49	Joback Method
cpg	408.90	J/mol×K	843.20	Joback Method
dvisc	0.0001846	Paxs	615.55	Joback Method

dvisc	0.0002306	Paxs	574.78	Joback Method
dvisc	0.0002981	Paxs	534.01	Joback Method
dvisc	0.0004019	Paxs	493.24	Joback Method
dvisc	0.0005720	Paxs	452.46	Joback Method
dvisc	0.0008731	Paxs	411.69	Joback Method
dvisc	0.0014623	Paxs	370.92	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U307585&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U307585&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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