

# 2-chloroethyl pentadecanoate

**Inchi:** InChI=1S/C17H33ClO2/c1-2-3-4-5-6-7-8-9-10-11-12-13-14-17(19)20-16-15-18/h2-16H2,  
**InchiKey:** MEQQQMOVIIYVGPT-UHFFFAOYSA-N  
**Formula:** C17H33ClO2  
**SMILES:** CCCCCCCCCCCCCC(=O)OCCCl  
**Mol. weight [g/mol]:** 304.90

## Physical Properties

Property code	Value	Unit	Source
gf	-153.59	kJ/mol	Joback Method
hf	-654.75	kJ/mol	Joback Method
hfus	46.77	kJ/mol	Joback Method
hvap	66.98	kJ/mol	Joback Method
log10ws	-5.95		Crippen Method
logp	5.860		Crippen Method
mcvol	270.070	ml/mol	McGowan Method
pc	1238.96	kPa	Joback Method
ripol	2105.00		NIST Webbook
ripol	2094.00		NIST Webbook
ripol	2107.00		NIST Webbook
ripol	2102.00		NIST Webbook
ripol	2094.00		NIST Webbook
ripol	2105.00		NIST Webbook
ripol	2599.00		NIST Webbook
ripol	2584.00		NIST Webbook
ripol	2571.00		NIST Webbook
ripol	2584.00		NIST Webbook
ripol	2591.00		NIST Webbook
ripol	2590.00		NIST Webbook
ripol	2572.00		NIST Webbook
ripol	2571.00		NIST Webbook
tb	702.08	K	Joback Method
tc	874.86	K	Joback Method
tf	383.43	K	Joback Method
vc	1.060	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	767.38	J/molxK	702.08	Joback Method
cpg	847.21	J/molxK	846.07	Joback Method
cpg	832.80	J/molxK	817.27	Joback Method
cpg	817.63	J/molxK	788.47	Joback Method
cpg	801.69	J/molxK	759.67	Joback Method
cpg	784.94	J/molxK	730.88	Joback Method
cpg	860.88	J/molxK	874.86	Joback Method
dvisc	0.0000966	Paxs	702.08	Joback Method
dvisc	0.0001285	Paxs	648.97	Joback Method
dvisc	0.0001799	Paxs	595.86	Joback Method
dvisc	0.0002691	Paxs	542.75	Joback Method
dvisc	0.0004391	Paxs	489.65	Joback Method
dvisc	0.0008072	Paxs	436.54	Joback Method
dvisc	0.0017567	Paxs	383.43	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R30862&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R30862&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient

<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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