

«beta»-Alanine, N-(1-naphthoyl)-, ethyl ester

Inchi: InChI=1S/C16H17NO3/c1-2-20-15(18)10-11-17-16(19)14-9-5-7-12-6-3-4-8-13(12)14/h3-9
InchiKey: SLYQVPOOQAIFIQ-UHFFFAOYSA-N
Formula: C16H17NO3
SMILES: CCOC(=O)CCNC(=O)c1cccc2ccccc12
Mol. weight [g/mol]: 271.31

Physical Properties

Property code	Value	Unit	Source
gf	19.82	kJ/mol	Joback Method
hf	-261.35	kJ/mol	Joback Method
hfus	37.35	kJ/mol	Joback Method
hvap	78.13	kJ/mol	Joback Method
log10ws	-4.16		Crippen Method
logp	2.523		Crippen Method
mvol	212.070	ml/mol	McGowan Method
pc	2331.52	kPa	Joback Method
rinpol	2448.00		NIST Webbook
rinpol	2448.00		NIST Webbook
tb	796.45	K	Joback Method
tc	1019.72	K	Joback Method
tf	516.47	K	Joback Method
vc	0.810	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	605.11	J/mol×K	796.45	Joback Method
cpg	618.23	J/mol×K	833.66	Joback Method
cpg	630.38	J/mol×K	870.87	Joback Method
cpg	641.62	J/mol×K	908.09	Joback Method
cpg	652.01	J/mol×K	945.30	Joback Method
cpg	661.61	J/mol×K	982.51	Joback Method
cpg	670.48	J/mol×K	1019.72	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U321944&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
h vap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
r in pol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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