

# Methyl 2,3,3-trichloropropanoate

<b>Other names:</b>	Propanoic acid, 2,3,3-trichloro, methyl ester
<b>Inchi:</b>	InChI=1S/C4H5Cl3O2/c1-9-4(8)2(5)3(6)7/h2-3H,1H3
<b>InchiKey:</b>	VCJSVVVJNDJVPM-UHFFFAOYSA-N
<b>Formula:</b>	C4H5Cl3O2
<b>SMILES:</b>	COC(=O)C(Cl)C(Cl)Cl
<b>Mol. weight [g/mol]:</b>	191.44
<b>CAS:</b>	20618-07-9

## Physical Properties

Property code	Value	Unit	Source
gf	-291.79	kJ/mol	Joback Method
hf	-428.47	kJ/mol	Joback Method
hfus	14.45	kJ/mol	Joback Method
hvap	46.03	kJ/mol	Joback Method
log10ws	-1.54		Crippen Method
logp	1.571		Crippen Method
mcvol	111.380	ml/mol	McGowan Method
pc	3668.65	kPa	Joback Method
rinpol	1031.00		NIST Webbook
rinpol	1029.00		NIST Webbook
rinpol	1031.00		NIST Webbook
rinpol	1027.00		NIST Webbook
rinpol	1031.00		NIST Webbook
tb	478.62	K	Joback Method
tc	688.86	K	Joback Method
tf	266.76	K	Joback Method
vc	0.418	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	192.56	J/molxK	478.62	Joback Method
cpg	223.09	J/molxK	653.82	Joback Method
cpg	217.64	J/molxK	618.78	Joback Method

cpg	211.87	J/molxK	583.74	Joback Method
cpg	205.76	J/molxK	548.70	Joback Method
cpg	199.32	J/molxK	513.66	Joback Method
cpg	228.21	J/molxK	688.86	Joback Method
dvisc	0.0003244	Paxs	478.62	Joback Method
dvisc	0.0004265	Paxs	443.31	Joback Method
dvisc	0.0005881	Paxs	408.00	Joback Method
dvisc	0.0008616	Paxs	372.69	Joback Method
dvisc	0.0013676	Paxs	337.38	Joback Method
dvisc	0.0024180	Paxs	302.07	Joback Method
dvisc	0.0049718	Paxs	266.76	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C20618079&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C20618079&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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