

DL-Alanine, N-methyl-N-(3-chloropropoxycarbonyl)-, hexyl

Inchi:
ester

InChI=1S/C14H26ClNO4/c1-4-5-6-7-10-19-13(17)12(2)16(3)14(18)20-11-8-9-15/h12H,4-

InchiKey:

ALVHZDRXLQHAQG-UHFFFAOYSA-N

Formula:

C14H26ClNO4

SMILES:

CCCCCOC(=O)C(C)N(C)C(=O)OCCCCI

Mol. weight [g/mol]:

307.81

Physical Properties

Property code	Value	Unit	Source
gf	-304.43	kJ/mol	Joback Method
hf	-775.38	kJ/mol	Joback Method
hfus	41.28	kJ/mol	Joback Method
hvap	71.11	kJ/mol	Joback Method
log10ws	-3.22		Crippen Method
logp	3.196		Crippen Method
mcvol	245.220	ml/mol	McGowan Method
pc	1598.72	kPa	Joback Method
rinpol	2010.00		NIST Webbook
rinpol	2010.00		NIST Webbook
tb	721.73	K	Joback Method
tc	903.95	K	Joback Method
tf	439.25	K	Joback Method
vc	0.928	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	693.57	J/mol×K	721.73	Joback Method
cpg	708.82	J/mol×K	752.10	Joback Method
cpg	723.24	J/mol×K	782.47	Joback Method
cpg	736.85	J/mol×K	812.84	Joback Method
cpg	749.66	J/mol×K	843.21	Joback Method
cpg	761.67	J/mol×K	873.58	Joback Method
cpg	772.91	J/mol×K	903.95	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U392778&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
h vap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
r in pol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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