

# Benzamide, 2,3,4-trifluoro-N-(hept-2-yl)-

<b>Inchi:</b>	InChI=1S/C14H18F3NO/c1-3-4-5-6-9(2)18-14(19)10-7-8-11(15)13(17)12(10)16/h7-9H,3-
<b>InchiKey:</b>	RVRDVFRBRFJVEY-UHFFFAOYSA-N
<b>Formula:</b>	C14H18F3NO
<b>SMILES:</b>	CCCCC(C)NC(=O)c1ccc(F)c(F)c1F
<b>Mol. weight [g/mol]:</b>	273.29

## Physical Properties

Property code	Value	Unit	Source
gf	-475.88	kJ/mol	Joback Method
hf	-782.89	kJ/mol	Joback Method
hfus	37.30	kJ/mol	Joback Method
hvap	61.36	kJ/mol	Joback Method
log10ws	-5.43		Crippen Method
logp	3.803		Crippen Method
mcvol	201.220	ml/mol	McGowan Method
pc	1846.75	kPa	Joback Method
rinpol	1709.00		NIST Webbook
rinpol	1709.00		NIST Webbook
tb	662.75	K	Joback Method
tc	848.05	K	Joback Method
tf	400.88	K	Joback Method
vc	0.800	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	545.12	J/mol×K	662.75	Joback Method
cpg	559.18	J/mol×K	693.63	Joback Method
cpg	572.49	J/mol×K	724.52	Joback Method
cpg	585.07	J/mol×K	755.40	Joback Method
cpg	596.94	J/mol×K	786.28	Joback Method
cpg	608.14	J/mol×K	817.17	Joback Method
cpg	618.67	J/mol×K	848.05	Joback Method

# Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U407263&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U407263&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvp:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinp:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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