

# 5,6-Dihydro-5-methyluracil

<b>Other names:</b>	5,6-Dihydro-2,4-dihydroxy-5-methylpyrimidine Dihydrothymine 2,4(1H,3H)-Pyrimidinedione, dihydro-5-methyl- Hydrouracil, 5-methyl- 5-Methyl-5,6-dihydrouracil 5,6-Dihydrothymine
<b>Inchi:</b>	InChI=1S/C5H8N2O2/c1-3-2-6-5(9)7-4(3)8/h3H,2H2,1H3,(H2,6,7,8,9)
<b>InchiKey:</b>	NBAKTGXDIBVZOO-UHFFFAOYSA-N
<b>Formula:</b>	C5H8N2O2
<b>SMILES:</b>	CC1CNC(=O)NC1=O
<b>Mol. weight [g/mol]:</b>	128.13
<b>CAS:</b>	696-04-8

## Physical Properties

Property code	Value	Unit	Source
gf	-54.09	kJ/mol	Joback Method
hf	-291.99	kJ/mol	Joback Method
hfus	18.74	kJ/mol	Joback Method
hvap	49.16	kJ/mol	Joback Method
log10ws	-0.48		Crippen Method
logp	-0.538		Crippen Method
mcvol	93.550	ml/mol	McGowan Method
pc	5304.68	kPa	Joback Method
rinpol	1523.80		NIST Webbook
rinpol	1523.80		NIST Webbook
tb	566.09	K	Joback Method
tc	826.71	K	Joback Method
tf	499.99	K	Joback Method
vc	0.337	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	221.19	J/molxK	566.09	Joback Method

cpg	235.27	J/mol×K	609.53	Joback Method
cpg	248.80	J/mol×K	652.96	Joback Method
cpg	261.65	J/mol×K	696.40	Joback Method
cpg	273.71	J/mol×K	739.84	Joback Method
cpg	284.87	J/mol×K	783.28	Joback Method
cpg	295.01	J/mol×K	826.71	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C696048&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C696048&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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