

# Succinic acid, 2,3-dichlorophenyl trans-4-methylcyclohexyl ester

<b>Inchi:</b>	InChI=1S/C17H20Cl2O4/c1-11-5-7-12(8-6-11)22-15(20)9-10-16(21)23-14-4-2-3-13(18)1
<b>InchiKey:</b>	LAAJLVGJKSMGKB-UHFFFAOYSA-N
<b>Formula:</b>	C17H20Cl2O4
<b>SMILES:</b>	CC1CCC(OC(=O)CCC(=O)Oc2cccc(Cl)c2Cl)CC1
<b>Mol. weight [g/mol]:</b>	359.24

## Physical Properties

Property code	Value	Unit	Source
gf	-289.55	kJ/mol	Joback Method
hf	-667.72	kJ/mol	Joback Method
hfus	39.92	kJ/mol	Joback Method
hvap	84.24	kJ/mol	Joback Method
log10ws	-5.55		Crippen Method
logp	4.801		Crippen Method
mcvol	255.130	ml/mol	McGowan Method
pc	1796.98	kPa	Joback Method
rinpol	2578.00		NIST Webbook
rinpol	2578.00		NIST Webbook
tb	867.32	K	Joback Method
tc	1099.89	K	Joback Method
tf	540.11	K	Joback Method
vc	0.958	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	752.80	J/molxK	867.32	Joback Method
cpg	808.73	J/molxK	1061.13	Joback Method
cpg	800.50	J/molxK	1022.37	Joback Method
cpg	790.80	J/molxK	983.61	Joback Method
cpg	779.63	J/molxK	944.84	Joback Method
cpg	766.97	J/molxK	906.08	Joback Method
cpg	815.51	J/molxK	1099.89	Joback Method
dvisc	0.0000853	Paxs	867.32	Joback Method

dvisc	0.0001059	Paxs	812.78	Joback Method
dvisc	0.0001357	Paxs	758.25	Joback Method
dvisc	0.0001806	Paxs	703.71	Joback Method
dvisc	0.0002522	Paxs	649.18	Joback Method
dvisc	0.0003745	Paxs	594.64	Joback Method
dvisc	0.0006022	Paxs	540.11	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U390073&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U390073&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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