

# Pentafluorobenzoic acid, 2-pentyl ester

<b>Other names:</b>	1-Methylbutyl pentafluorobenzoate Pentan-2-yl 2,3,4,5,6-pentafluorobenzoate
<b>Inchi:</b>	InChI=1S/C12H11F5O2/c1-3-4-5(2)19-12(18)6-7(13)9(15)11(17)10(16)8(6)14/h5H,3-4H2
<b>InchiKey:</b>	BKCNCJFSIBDJLT-UHFFFAOYSA-N
<b>Formula:</b>	C12H11F5O2
<b>SMILES:</b>	CCCC(C)OC(=O)c1c(F)c(F)c(F)c(F)c1F
<b>Mol. weight [g/mol]:</b>	282.21
<b>CAS:</b>	99483-15-5

## Physical Properties

Property code	Value	Unit	Source
gf	-1095.99	kJ/mol	Joback Method
hf	-1342.46	kJ/mol	Joback Method
hfus	33.60	kJ/mol	Joback Method
hvap	52.58	kJ/mol	Joback Method
log10ws	-5.16		Crippen Method
logp	3.728		Crippen Method
mcvol	172.470	ml/mol	McGowan Method
pc	1915.26	kPa	Joback Method
ripol	1247.00		NIST Webbook
ripol	1277.00		NIST Webbook
ripol	1255.00		NIST Webbook
ripol	1277.00		NIST Webbook
ripol	1253.00		NIST Webbook
ripol	1257.00		NIST Webbook
ripol	1253.00		NIST Webbook
ripol	1506.00		NIST Webbook
ripol	1518.00		NIST Webbook
ripol	1518.00		NIST Webbook
ripol	1483.00		NIST Webbook
ripol	1506.00		NIST Webbook
ripol	1498.00		NIST Webbook
tb	597.74	K	Joback Method
tc	771.09	K	Joback Method
tf	374.13	K	Joback Method
vc	0.708	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	433.19	J/mol×K	597.74	Joback Method
cpg	444.56	J/mol×K	626.63	Joback Method
cpg	455.45	J/mol×K	655.52	Joback Method
cpg	465.86	J/mol×K	684.41	Joback Method
cpg	475.78	J/mol×K	713.31	Joback Method
cpg	485.23	J/mol×K	742.20	Joback Method
cpg	494.19	J/mol×K	771.09	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C99483155&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C99483155&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpolar:</b>	Non-polar retention indices
<b>ripolar:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

**tf:** Normal melting (fusion) point

**vc:** Critical Volume

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