

Diglycolic acid, 2,6-dichlorophenyl ethyl ester

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| Inchi: | InChI=1S/C12H12Cl2O5/c1-2-18-10(15)6-17-7-11(16)19-12-8(13)4-3-5-9(12)14/h3-5H,2 |
| InchiKey: | YNWXYMRLGXKSPS-UHFFFAOYSA-N |
| Formula: | C12H12Cl2O5 |
| SMILES: | CCOC(=O)COCC(=O)Oc1c(Cl)cccc1Cl |
| Mol. weight [g/mol]: | 307.13 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|----------------------|----------------|
| gf | -453.39 | kJ/mol | Joback Method |
| hf | -730.72 | kJ/mol | Joback Method |
| hfus | 35.25 | kJ/mol | Joback Method |
| hvap | 75.40 | kJ/mol | Joback Method |
| log10ws | -2.78 | | Crippen Method |
| logp | 2.478 | | Crippen Method |
| mvol | 201.410 | ml/mol | McGowan Method |
| pc | 2338.29 | kPa | Joback Method |
| rinpol | 2479.00 | | NIST Webbook |
| rinpol | 2479.00 | | NIST Webbook |
| tb | 760.46 | K | Joback Method |
| tc | 977.36 | K | Joback Method |
| tf | 502.85 | K | Joback Method |
| vc | 0.763 | m ³ /kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|-----------|---------|-----------------|---------------|
| cpg | 507.79 | J/molxK | 760.46 | Joback Method |
| cpg | 553.85 | J/molxK | 941.21 | Joback Method |
| cpg | 546.45 | J/molxK | 905.06 | Joback Method |
| cpg | 538.12 | J/molxK | 868.91 | Joback Method |
| cpg | 528.89 | J/molxK | 832.76 | Joback Method |
| cpg | 518.77 | J/molxK | 796.61 | Joback Method |
| cpg | 560.32 | J/molxK | 977.36 | Joback Method |
| dvisc | 0.0000948 | Paxs | 760.46 | Joback Method |

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|-------|-----------|------|--------|---------------|
| dvisc | 0.0001161 | Paxs | 717.53 | Joback Method |
| dvisc | 0.0001458 | Paxs | 674.59 | Joback Method |
| dvisc | 0.0001890 | Paxs | 631.65 | Joback Method |
| dvisc | 0.0002543 | Paxs | 588.72 | Joback Method |
| dvisc | 0.0003586 | Paxs | 545.79 | Joback Method |
| dvisc | 0.0005363 | Paxs | 502.85 | Joback Method |

Sources

| | |
|------------------------|---|
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=U382295&Units=SI |
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci9903071 |
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |

Legend

| | |
|----------------------------|---|
| cp_g: | Ideal gas heat capacity |
| dvisc: | Dynamic viscosity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| log₁₀ws: | Log ₁₀ of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| m_{cvol}: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| rinpol: | Non-polar retention indices |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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