

Glutaric acid, 3-chlorophenyl 2-hexyl ester

| | |
|-----------------------------|--|
| Inchi: | InChI=1S/C17H23ClO4/c1-3-4-7-13(2)21-16(19)10-6-11-17(20)22-15-9-5-8-14(18)12-15 |
| InchiKey: | GZHWUSQDJUYJLJ-UHFFFAOYSA-N |
| Formula: | C17H23ClO4 |
| SMILES: | CCCCC(C)OC(=O)CCCC(=O)Oc1cccc(Cl)c1 |
| Mol. weight [g/mol]: | 326.81 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|----------------------|----------------|
| gf | -287.17 | kJ/mol | Joback Method |
| hf | -679.77 | kJ/mol | Joback Method |
| hfus | 39.69 | kJ/mol | Joback Method |
| hvap | 78.68 | kJ/mol | Joback Method |
| log10ws | -5.21 | | Crippen Method |
| logp | 4.538 | | Crippen Method |
| mcvol | 253.750 | ml/mol | McGowan Method |
| pc | 1637.78 | kPa | Joback Method |
| rinpol | 2231.00 | | NIST Webbook |
| rinpol | 2231.00 | | NIST Webbook |
| tb | 809.59 | K | Joback Method |
| tc | 1016.62 | K | Joback Method |
| tf | 479.53 | K | Joback Method |
| vc | 0.971 | m ³ /kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|-----------|---------|-----------------|---------------|
| cpg | 731.89 | J/molxK | 809.59 | Joback Method |
| cpg | 746.34 | J/molxK | 844.09 | Joback Method |
| cpg | 759.73 | J/molxK | 878.60 | Joback Method |
| cpg | 772.07 | J/molxK | 913.10 | Joback Method |
| cpg | 783.39 | J/molxK | 947.61 | Joback Method |
| cpg | 793.70 | J/molxK | 982.11 | Joback Method |
| cpg | 803.02 | J/molxK | 1016.62 | Joback Method |
| dvisc | 0.0007413 | Paxs | 479.53 | Joback Method |

| | | | | |
|-------|-----------|------|--------|---------------|
| dvisc | 0.0004031 | Paxs | 534.54 | Joback Method |
| dvisc | 0.0002455 | Paxs | 589.55 | Joback Method |
| dvisc | 0.0001628 | Paxs | 644.56 | Joback Method |
| dvisc | 0.0001151 | Paxs | 699.57 | Joback Method |
| dvisc | 0.0000857 | Paxs | 754.58 | Joback Method |
| dvisc | 0.0000663 | Paxs | 809.59 | Joback Method |

Sources

| | |
|------------------------|---|
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=U393603&Units=SI |
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci990307l |
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |

Legend

| | |
|-----------------|---|
| cpg: | Ideal gas heat capacity |
| dvisc: | Dynamic viscosity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| rinpol: | Non-polar retention indices |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

Latest version available from:

<https://www.chemeo.com/cid/114-008-6/Glutaric-acid-3-chlorophenyl-2-hexyl-ester.pdf>

Generated by Cheméo on 2024-04-27 20:44:26.796915098 +0000 UTC m=+16539915.717492443.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.