

Glutaric acid, 2,2-dichloroethyl 2,4,5-trichlorophenyl ester

Inchi: InChI=1S/C13H11Cl5O4/c14-7-4-9(16)10(5-8(7)15)22-13(20)3-1-2-12(19)21-6-11(17)18/
InchiKey: LSFWTKYGIGZMPJ-UHFFFAOYSA-N
Formula: C13H11Cl5O4
SMILES: O=C(CCCC(=O)Oc1cc(Cl)c(Cl)cc1Cl)OCC(Cl)Cl
Mol. weight [g/mol]: 408.49

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|----------------------|----------------|
| gf | -387.83 | kJ/mol | Joback Method |
| hf | -683.11 | kJ/mol | Joback Method |
| hfus | 45.34 | kJ/mol | Joback Method |
| hvap | 88.64 | kJ/mol | Joback Method |
| log10ws | -5.71 | | Crippen Method |
| logp | 5.069 | | Crippen Method |
| mvol | 246.350 | ml/mol | McGowan Method |
| pc | 1949.23 | kPa | Joback Method |
| rinpol | 2660.00 | | NIST Webbook |
| rinpol | 2660.00 | | NIST Webbook |
| tb | 877.75 | K | Joback Method |
| tc | 1107.09 | K | Joback Method |
| tf | 579.17 | K | Joback Method |
| vc | 0.943 | m ³ /kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|-----------|---------|-----------------|---------------|
| cpg | 595.61 | J/molxK | 877.75 | Joback Method |
| cpg | 604.07 | J/molxK | 915.97 | Joback Method |
| cpg | 611.57 | J/molxK | 954.20 | Joback Method |
| cpg | 618.11 | J/molxK | 992.42 | Joback Method |
| cpg | 623.70 | J/molxK | 1030.64 | Joback Method |
| cpg | 628.34 | J/molxK | 1068.87 | Joback Method |
| cpg | 632.03 | J/molxK | 1107.09 | Joback Method |
| dvisc | 0.0003942 | Paxs | 579.17 | Joback Method |

| | | | | |
|-------|-----------|------|--------|---------------|
| dvisc | 0.0002600 | Paxs | 628.93 | Joback Method |
| dvisc | 0.0001823 | Paxs | 678.70 | Joback Method |
| dvisc | 0.0001342 | Paxs | 728.46 | Joback Method |
| dvisc | 0.0001027 | Paxs | 778.22 | Joback Method |
| dvisc | 0.0000812 | Paxs | 827.99 | Joback Method |
| dvisc | 0.0000659 | Paxs | 877.75 | Joback Method |

Sources

| | |
|------------------------|---|
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=U392160&Units=SI |
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci9903071 |
| Crippen Method: | https://www.cheméo.com/doc/models/crippen_log10ws |
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |

Legend

| | |
|---------------------------------------|---|
| cp_g: | Ideal gas heat capacity |
| dvisc: | Dynamic viscosity |
| g_f: | Standard Gibbs free energy of formation |
| h_f: | Enthalpy of formation at standard conditions |
| h_{fus}: | Enthalpy of fusion at standard conditions |
| h_{vap}: | Enthalpy of vaporization at standard conditions |
| log₁₀w_s: | Log ₁₀ of Water solubility in mol/l |
| log_p: | Octanol/Water partition coefficient |
| mc_{vol}: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| rin_{pol}: | Non-polar retention indices |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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