

# Carbonochloridic acid, 2,2,2-trichloroethyl ester

<b>Other names:</b>	Formic acid, chloro-, 2,2,2-trichloroethyl ester Trichloroethyl chloroformate 2,2,2-Trichloroethoxycarbonyl chloride 2,2,2-Trichloroethyl chloroformate Chloroformic acid 2,2,2-trichloroethyl ester
<b>Inchi:</b>	InChI=1S/C3H2Cl4O2/c4-2(8)9-1-3(5,6)7/h1H2
<b>InchiKey:</b>	LJCZNYWLQZZIOS-UHFFFAOYSA-N
<b>Formula:</b>	C3H2Cl4O2
<b>SMILES:</b>	O=C(Cl)OCC(Cl)(Cl)Cl
<b>Mol. weight [g/mol]:</b>	211.86
<b>CAS:</b>	17341-93-4

## Physical Properties

Property code	Value	Unit	Source
gf	-304.42	kJ/mol	Joback Method
hf	-421.76	kJ/mol	Joback Method
h <sub>fus</sub>	15.69	kJ/mol	Joback Method
h <sub>vap</sub>	47.67	kJ/mol	Joback Method
log <sub>10</sub> ws	-2.63		Crippen Method
logp	2.732		Crippen Method
m <sub>cvol</sub>	109.530	ml/mol	McGowan Method
pc	3960.52	kPa	Joback Method
tb	444.70	K	NIST Webbook
tc	714.30	K	Joback Method
tf	317.83	K	Joback Method
vc	0.412	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
c <sub>pg</sub>	176.92	J/mol×K	490.82	Joback Method
c <sub>pg</sub>	198.16	J/mol×K	677.05	Joback Method
c <sub>pg</sub>	194.68	J/mol×K	639.81	Joback Method
c <sub>pg</sub>	190.84	J/mol×K	602.56	Joback Method

cpg	186.61	J/molxK	565.31	Joback Method
cpg	181.98	J/molxK	528.07	Joback Method
cpg	201.29	J/molxK	714.30	Joback Method
dvisc	0.0003762	Paxs	490.82	Joback Method
dvisc	0.0004782	Paxs	461.99	Joback Method
dvisc	0.0006276	Paxs	433.16	Joback Method
dvisc	0.0008562	Paxs	404.33	Joback Method
dvisc	0.0012252	Paxs	375.49	Joback Method
dvisc	0.0018609	Paxs	346.66	Joback Method
dvisc	0.0030492	Paxs	317.83	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C17341934&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C17341934&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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